



RESERVOIR ENGINEERING GRADUATE CERTIFICATE - Week 8

Drive Mechanisms -EOR

A special course by IFP Training for REPSOL ALGERIA Alger –December 18 to 22, 2016



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An IFP Training Course for REPSOL

Material balance MBALTM software

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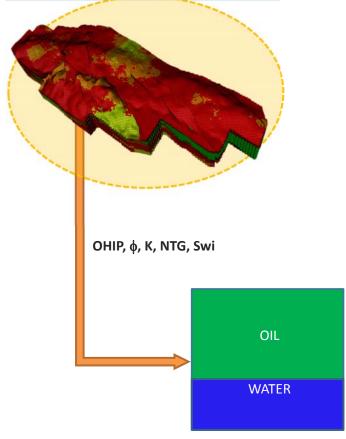


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Tank model



THE RESERVOIR IS REPRESENTED BY A TANK WITH A UNIQUE POROSITY, PRESSURE, SATURATION

- Very simple model
- ► Material balance equation
- ► Checking data consistency
- ▶ Determines the fluids in place
- Estimates the aquifer type and size
- Production prediction

Tank model

- ▶ The model is set up by entering the fluid properties (ρ, μ, FVF and the reservoir parameters (geometry, φ, K, Swi). The calculations considers viscous and gravity forces.
- ► The Buckley-Leveret technique is used to calculate the movement of a flood front along a 1D reservoir section. Water displacing oil from downdip and gas displaying oil from the updip direction can be modeled.
- ► From PVT and reservoir properties, fractional flow can be calculated for either gas or water displacing oil. The effect of dip angle, injection rates and relative permeability can be evaluated.

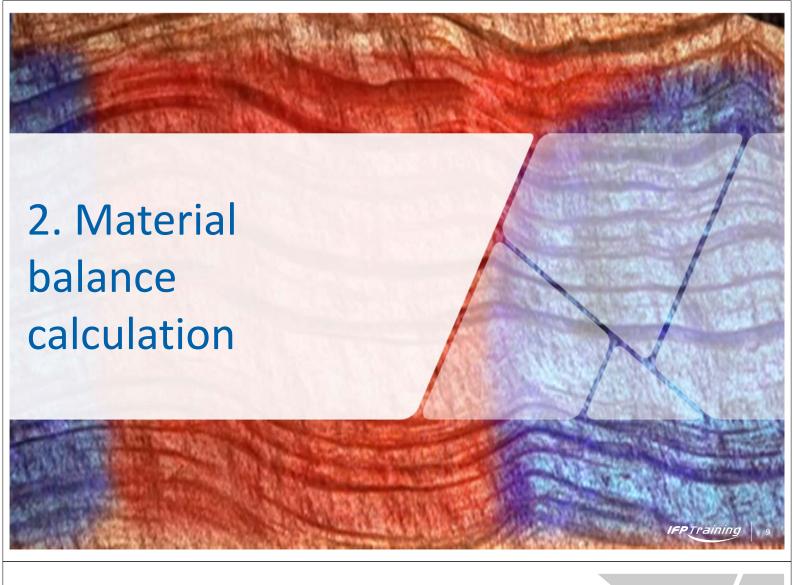


Tank model

- ► The material balance is suggested as a necessary step prior to carrying out a simulation study.
- ▶ The material balance will always enable the drive mechanisms to be identified.
- ► A material balance study can provide OHIP and drive mechanisms as inputs to the simulation.

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Symbols used in material balance equations

Bg	Formation volume factor for gas (res.vol./st.vol.)
Во	Formation volume factor for oil (res.vol./st.vol.)
Bw	Formation volume factor for water (res.vol./st.vol.)

Cr, Cp Pore compressibility (pressure-1)
Cw Water compressibility (pressure-1)

ΔP P2-P1

Ef,w Rock and water expansion/compression term

Eg Gas cap expansion term

Eo Oil & solution gas expansion term
G Original gas in place (st.vol.)
Gi Cumulative gas injected (st.vol.)
Gp Cumulative gas produced (st.vol.)

m Initial gas cap size (res.vol. of gas cap)/(res.vol. of oil zone)

N Original oil in place (st.vol.)
Np Cumulative oil produced (st.vol.)

P Pressure

Pb Bubble point Pressure

Rp Cumulative producing gas-oil ratio (st.vol./st.vol.) = Gp/Np

Rs Solution gas-oil ratio (st.vol. gas/st.vol. oil)

Sg Gas saturation
So Oil saturation
Sw Water saturation
T Temperature

Vb Bulk volume (res.vol.)
Vp Pore volume (res.vol.)

We Cumulative aquifer influx (st.vol.)
Wi Cumulative water injected (st.vol.)
Wp Cumulative water produced (st.vol.)

ρ Density (mass/vol.)

φ Porosity

Natural drainage mechanisms (primary recovery)

- Rock & fluid expansion
- Solution gas drive
- Initial gas cap expansion
- Aquifer water influx
- Combination drive
- Gravity drainage

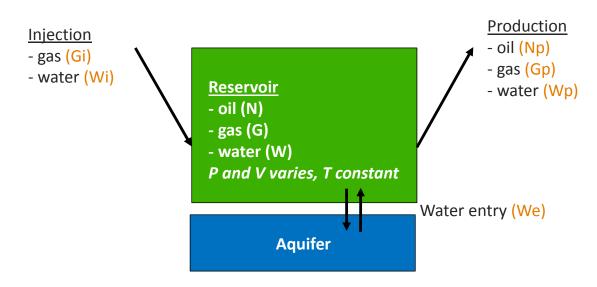
Secondary recovery

- Water injection
- Immiscible gas injection



Remember... Principles of material balance calculation

▶ In material balance calculation, the reservoir can be seen as a black box (no heterogeneities), whose pore volume varies (pore shrinkage), where fluids go in and out, and whose pressure varies accordingly



N, G, W, Np, Gp, Wp, Gi, Wi are volumes, expressed in standard conditions

- ▶ The oil rim is produced and the production at the surface will consist of oil, gas and water
- ▶ The volumetric material balance expressed at reservoir conditions is:

Initial volume occupied by the oil =

oil volume left in the reservoir with its dissolved gas

- + liberated gas from oil and staying in the reservoir
- + gas volume from the initial gas cap invading the oil zone
- + water entry
- produced water



Generalized material balance

▶ The material balance equation becomes:

$$\begin{aligned} N.Bo_i &= (N-Np)Bo + [(N.Rs_i - Gp) - (N-Np)Rs]Bg \\ &+ m.N.Bo_i \left(\frac{Bg}{Bg_i} - 1\right) + We - Wp.Bw \end{aligned}$$

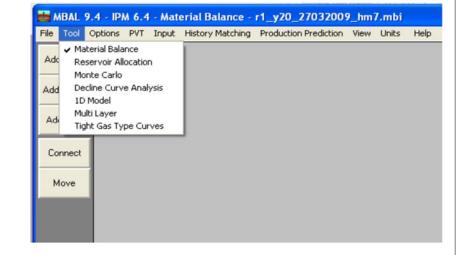
▶ Introducing Rp definition: $Rp = \frac{Gp}{Np}$

$$\begin{split} Np[Bo + (Rp - Rs)Bg] &= N[(Bo - Bo_i) + (Rs_i - Rs)Bg] \\ + m.N.Bo_i \left(\frac{Bg}{Bg_i} - 1\right) + We - Wp.Bw \end{split}$$



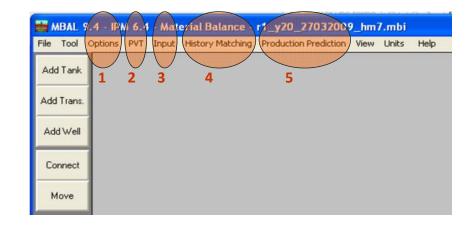
MBAL tools

- Material Balance
 - Production data matching
 - Production forecast
- Reservoir Allocation
- Monte Carlo volumetrics
- Decline curve Analysis
- ▶ 1-D model (Buckley Leverett)
- ► Multi Layer (relative permeability averaging)
- ▶ Tight Gas Type Curves



Start using MBAL

- Applications: Identify/confirm production mechanisms
 - Checking data consistency (OOIP, OGIP, Aquifer)
 - Production prediction
 - Sensitivities to different parameters
- 5 main modules:
 - 1. Options
 - 2. PVT
 - 3. Input
 - Well data
 - Tank data
 - Transmissibility data
 - 4. History matching
 - 5. Production prediction



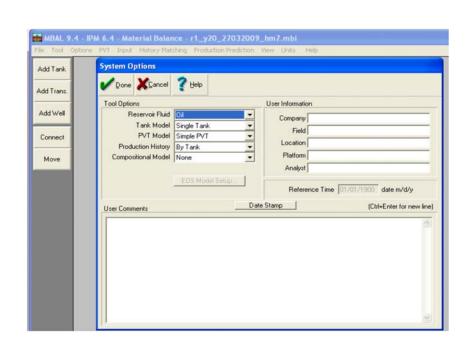


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Options

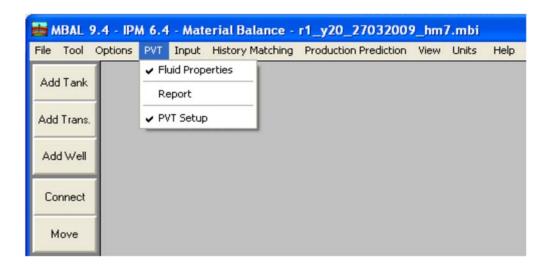
Definition of the model:

- Reservoir fluid:
 - Oil
 - Gas
 - Ret. Condensate
 - General
- Tank model:
 - Single Tank
 - Multiple Tanks
- PVT model:
 - Single PVT
 - Variable PVT
- Production history:
 - By Tank
 - By Well
- Compositional model:
 - None
 - Tracking
 - Fully Compositional



▶ Applications:

bring productions back to reservoir conditions





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PVT - possibilities

- ▶ Using "raw" correlations (1)
 - if few laboratory data (prospect)
- Using PVT tables data from PVT laboratory (2)
 - if PVT tables are coherent & complete
- Using matched correlations (3)
 - if PVT tables are not complete
- ▶ Using lab data & correlations (4)
 - ullet MBAL uses tables inside the pressure range & correlations outside ullet Recommended



MBAL – PVT quality

The quality of the Mbal model depends on the PVT quality

→use complete black oil tables (Bo, Rs, μo, Bg) from PVT package calculator with a wide range of Pressure & Temperature

For oil

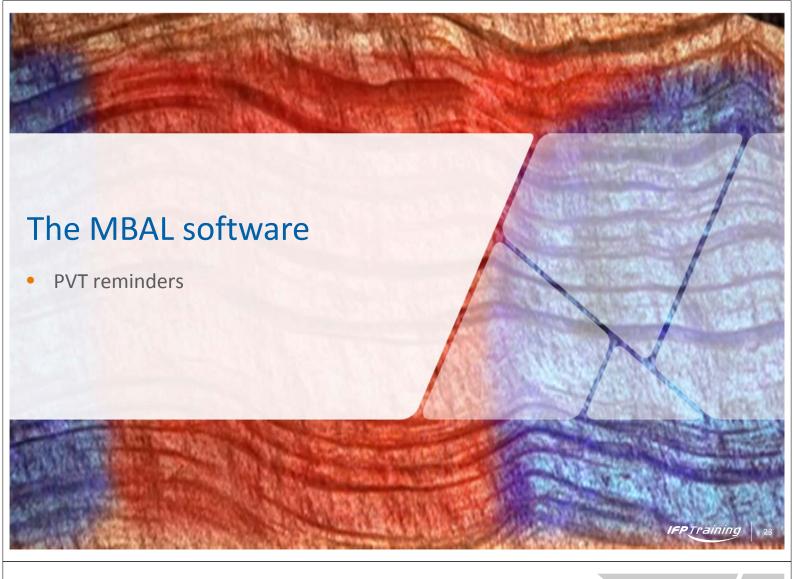
a composite PVT is necessary for oil

For condensate gas

- a CCE (Constant Composition Expansion) simulation is needed as input in MBAL
- a check of the liquid drop out deposit in the reservoir is imperative.

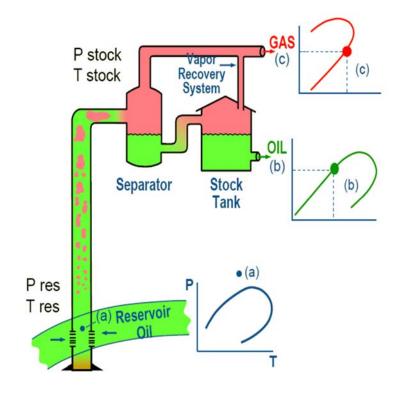
For dry or wet gas

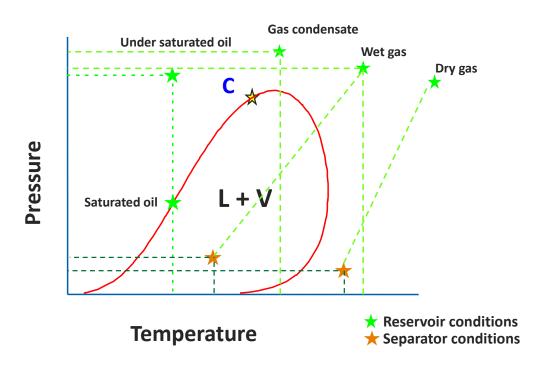
a composite PVT is necessary as input in MBAL



Fluid description

At reservoir conditions, the fluid is outside the phase envelop, fluids have different classifications depending on the position respect to the critical point. When reservoir fluids are produced, conditions change and we pass the bubble point or the dew point line (except for the dry gas) and we will have two phases at surface conditions.





BLACK OIL MODEL: small variations of composition

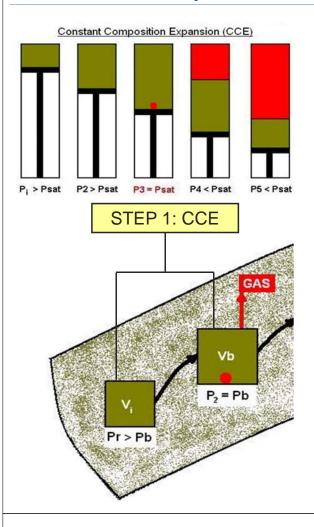
COMPOSITIONAL MODEL: recommended near the critical point



Reservoir fluid representation - Composite PVT

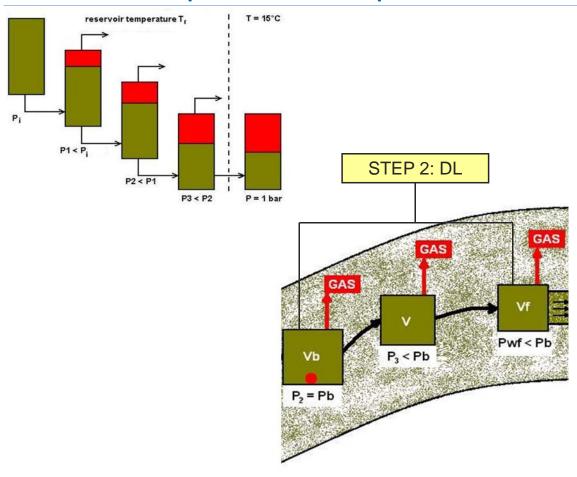
- ► A fluid sample will follow a composite path between its original location in the reservoir and its final destination at the surface
- First step
 - Fluid moves in the reservoir above saturation pressure
 - Volume change versus pressure is identical to CCE
- Second step
 - Fluid moves in the reservoir below saturation pressure
 - Liquid composition versus pressure is identical to a CVD
- Third step
 - Fluid has reached the well bore
 - Volume and composition surfaces are identical to a flash process

Reservoir fluid representation - Composite PVT

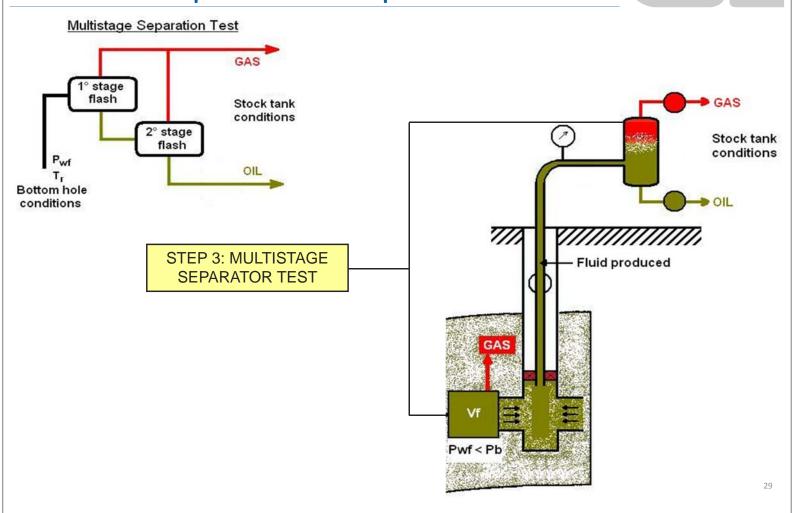


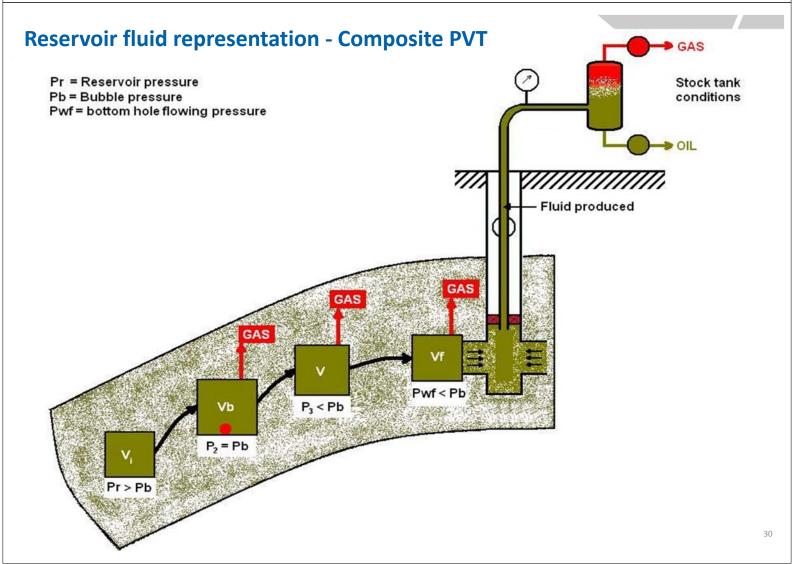


Reservoir fluid representation - Composite PVT



Reservoir fluid representation - Composite PVT





...Which PVT to choose?

► A fluid sample will follow a composite path between its original location in the reservoir and its final destination at the surface

First step

- Fluid moves in the reservoir above saturation pressure
- Volume change versus pressure is identical to CCE

Second step

- Fluid moves in the reservoir below saturation pressure
- Liquid composition versus pressure is identical to a CVD

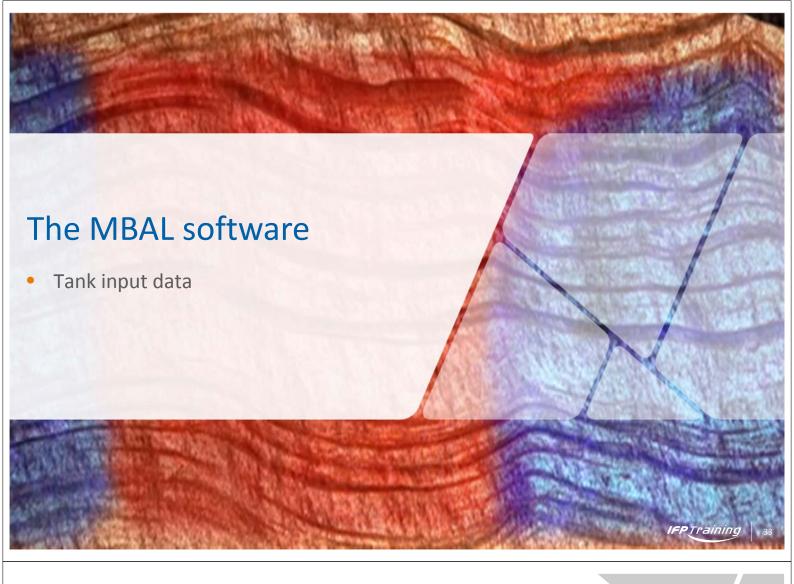
Third step

- Fluid has reached the well bore
- Volume and composition surfaces are identical to a flash process

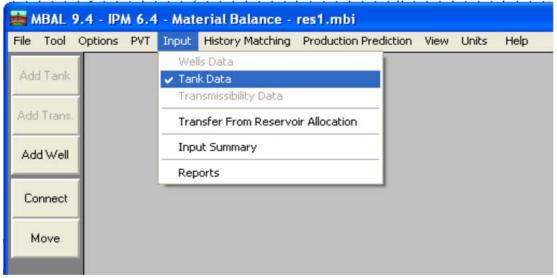


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Notes



Input



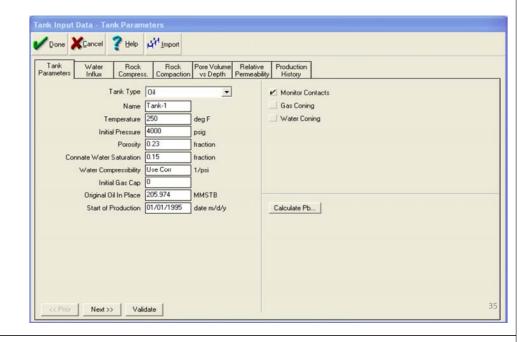
► Applications:

- Define well parameters
 - (if production history « by well » was selected in the Option module)
- Define tank parameters
- Define transmissibility parameters
 - (if multiple tanks was selected in the option module)

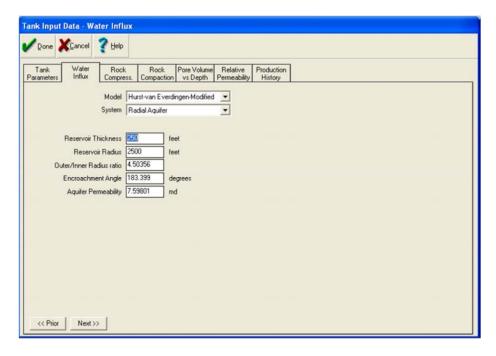
Input/Tank data – reservoir parameters

▶Tank/Reservoir characteristics:

- Tank type (Oil, water,...)
- Name
- Temperature
- Initial pressure
- Porosity
- Swc (connate water saturation)
- Water compressibility
- OOIP, OWIP
- Initial Gas cap via m
- Start of production

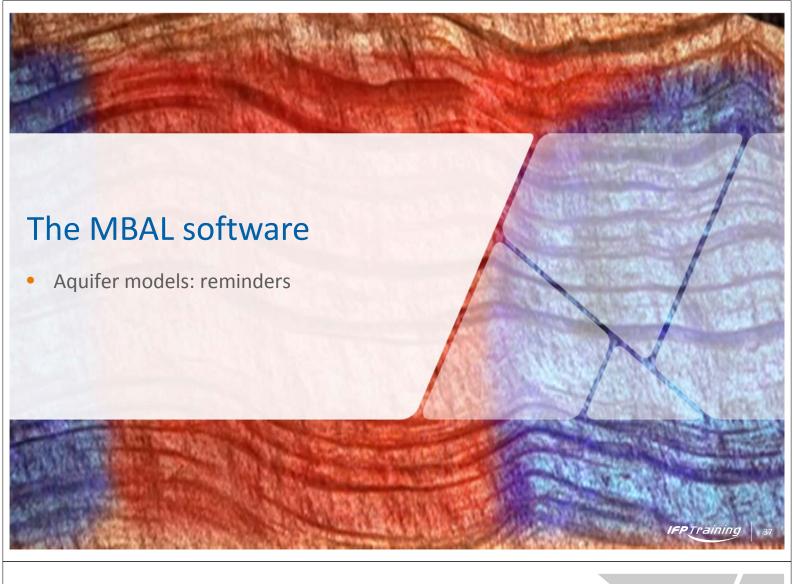


Input/Tank data – water influx



► Aquifer characteristics:

- Aquifer type
- Aquifer system (radial, linear, bottom,...)
- Aquifer geometric parameters



Water drive

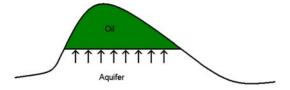
General principles

- ► The primary source of energy is provided by the water influx into the reservoir, which results in pressure maintenance
- ▶ In most cases, the energy comes from aquifer compressibility: Ca = Cp + Cw
- ► Water drive effectiveness is a function of the aquifer connection in the short term and the aquifer volume in long term

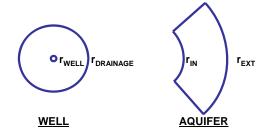
Water drive

Aquifer configuration

▶ The bottom water drive aguifer: it is in contact with the entire hydrocarbon area. The water invasion occurring vertically is governed by the reservoir vertical permeability.



▶ The edge water drive aquifer: the water entries take place laterally. Horizontal permeability is governing the water movement.





Natural water influx

Material balance

- Assuming P_h << P (for simplicity)</p>
 - Oil volume expands a.
 - b. Water volume expands
 - Pore volume decreases C.
 - Aquifer expands => Water entry W_e
 - e. Water production W_p
- ▶ Oil production = a + b + c + d e

$$Np.Bo = N.Bo_i.Ce(Pi - P) + We - Wp.Bw$$

$$RF = \frac{Np}{N} = \frac{Bo_i}{Bo}Ce(Pi - P) + \frac{We}{N.Bo}$$

▶ The water entry depends on the aquifer model, considering an instantaneous expansion:

$$We = Ca.Vw.(Pi - P)$$

 $Ca = Cw + Cp$

Natural water influx

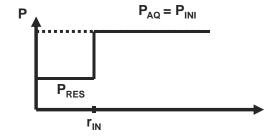
Water entry calculation

Aquifer models

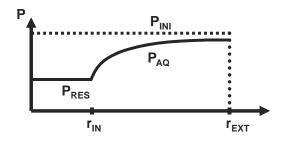
- **1.** Small pot: Independent time equation We = Ca.Vw.(Pi P)
- 2. Pseudo steady state: Fetkovich

$$\Delta We = J.(Pi - P).\Delta t$$

J is the Fetkovich productivity index



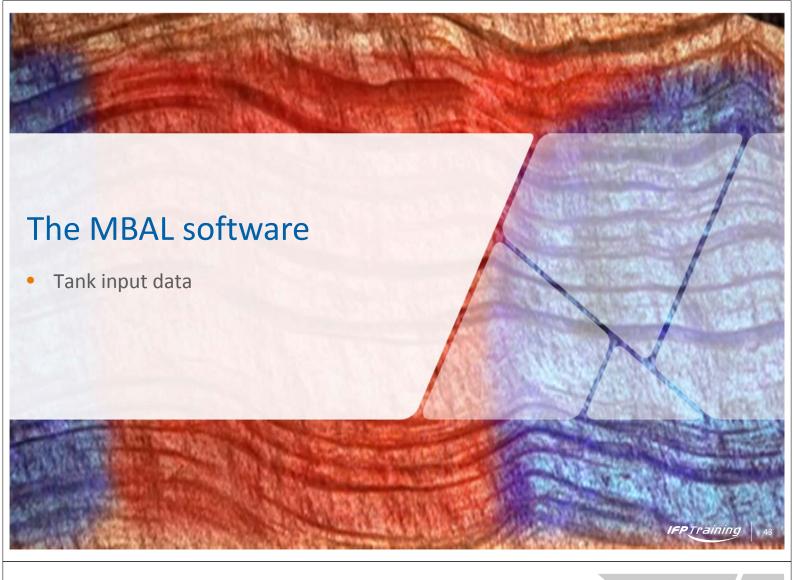
3. Transient: Carter Tracy model, approximate solution to diffusivity equation



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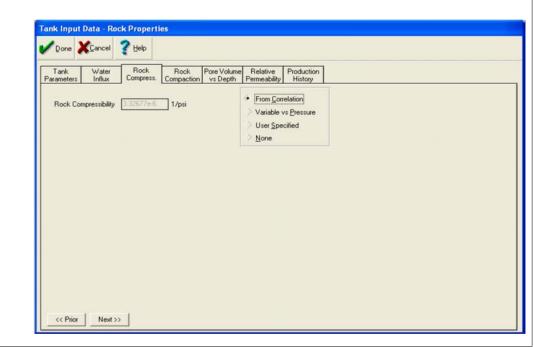
Notes



Input/Tank data – rock compressibility

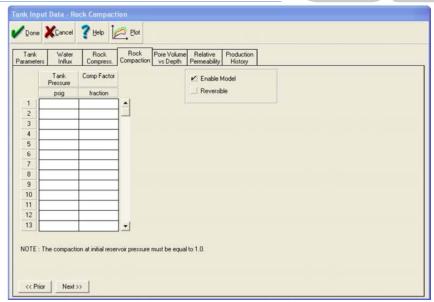
► Rock compressibility characteristics:

- Use of internal correlation
 - If the porosity of tank $\Phi > 0.3$ then Cf=3.2e⁻⁶
 - If the porosity of tank Φ < 0.3 then Cf=3.2e⁻⁶+(0.3- Φ)^{2.415}*7.8e⁻⁵
- Variable vs. Pressure:
- User specified
- None



 $C_f = -\frac{1}{V}\frac{dV}{dP} \cong -\frac{1}{V_i}\frac{V-V_i}{P-P_i}$

Input/Tank data – rock compaction



► Rock compaction characteristics:

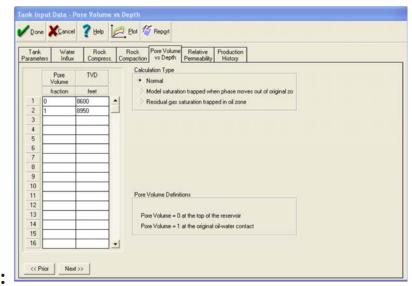
- Define a compaction factor of the pore volume vs. tank pressure
 - •PV = PVi x compaction factor(P)
- Define a reversible compaction factor of the pore volume vs. tank pressure
 - •To have a PV increase if the reservoir re-pressurizes

Be careful if you use a rock compaction model & a rock compressibility model then MBAL will calculate a PV by using: $PV = PVi \times (1-Cf(Pi-P)) \times compaction factor(P)$



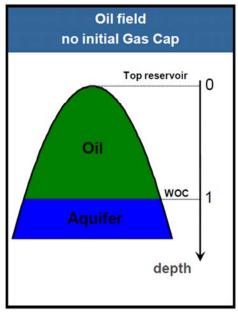
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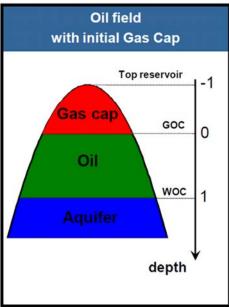
Input/Tank data - pore volume vs. depth

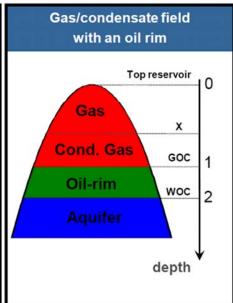


Pore volume vs. depth characteristics:

- Define a normal pore volume vs. depth law
- Define a saturation trapped when a phase moves out of the original zone:
 - When a phase invades the pore volume originally occupied by another phase, then a given saturation can be set as trapped
- Define a residual gas saturation trapped in oil zone
 - The gas will remain in the oil pore volume until the critical gas saturation is reached
- Necessary if the option 'Monitor contact' is switched 'on' ('tank parameters')





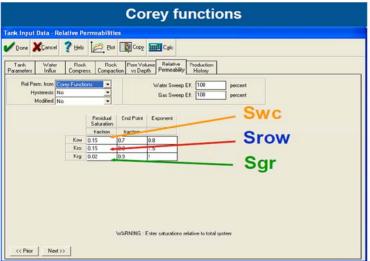


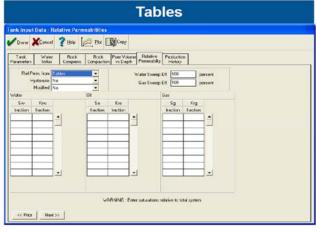


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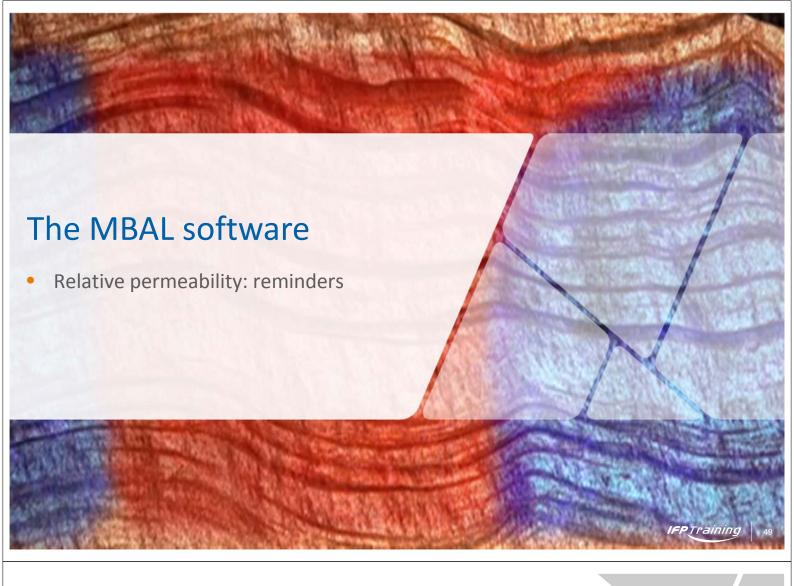
Input/Tank data - relative permeability

- 2 ways to introduce relative permeability curves:
 - By using Corey functions
 - By using tables





- for fluid contacts monitoring
- only for production prediction module



Relative permeability

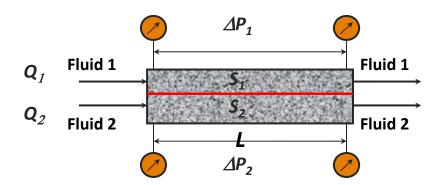
Darcy's law for multiphase flow

▶ In the case of diphasic flow

Monophasic Darcy's law is extended by introducing the effective permeability that takes
the presence of the other fluid in the porous medium into account

$$Q_1 = \frac{k_1}{\mu} A \frac{dP_1}{dx}$$
 where k_1 is the effective permeability of fluid 1 by respect to fluid 2

$$Q_2 = \frac{k_2}{\mu} A \frac{dP_{2w}}{dx}$$
 where k_2 is the effective permeability of fluid 2 by respect to fluid 1



Relative permeability

Definition of relative permeability

▶ From the previous extended Darcy's law for multiphasic flow

$$Q_1 = \frac{kk_{r1}}{\mu} A \frac{dP_1}{dx}$$

$$Q_2 = \frac{kk_{r2}}{\mu} A \frac{dP_{2w}}{dx}$$

where k_{r1} is the relative permeability of fluid 1 by respect to fluid 2 (adimensional)

k_{r2} is the relative permeability of fluid 2 by respect of fluid 1 (adimensional)

k is the monophasic permeability (Darcy)

Obviously, we get

$$k_{r1} = k_1/k$$

$$k_{r2} = k_2/k$$

Effective and relative permeabilities depend on the fluid saturations

$$k_{r1} = k_{r1}(S_1)$$

$$k_{r2} = k_{r2}(S_2)$$



Relative permeability

Definitions

Absolute permeability

Permeability of a rock completely saturated with one fluid: k_{qir} , k_{swi}

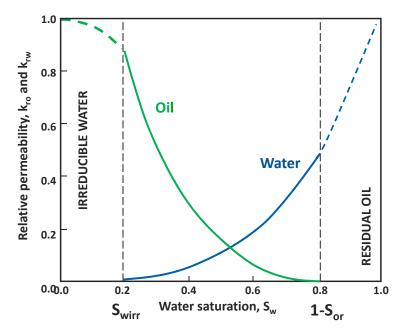
Effective permeability

- Permeability of a rock to one fluid when the rock is only partially saturated with that fluid: $k_o(S_w)$, $k_w(S_w)$
- Effective permeability is the measurement of the ability of the porous media to conduct one fluid in presence of the others

Relative permeability

Ratio of effective permeability to absolute permeability

Relative permeability curves - O/W system



 k_r curves **dictate the flow of fluids** in the reservoir and are used for production prediction in MBAL

END POINTS:

S_{wirr} irreducible water saturation

k_{ro,MAX} corresponds au maximum oil Kr (end point)

S_{or} residual oil saturation

→ k_{rw,MAX} corresponds au maximum oil Kr (end point)



Relative permeability - Corey normalization

Saturation normalization

$$S_{wn} = \frac{S_w - S_{wi}}{1 - S_{wi} - S_{or}}$$

normalized water saturation (for relative permeability)

$$S_{on} = 1 - S_{wn} = \frac{1 - S_{or} - S_w}{1 - S_{wi} - S_{or}}$$

normalized oil saturation

► Corey normalization for relative permeability

$$k_{rw} = k_{rwmax} S_{wn}^{nw}$$

$$k_{ro} = k_{romax} S_{on}^{no}$$

where:

 $k_{rwmax} = k_{rw}(S_{or})$

maximum relative permeability to water

 $k_{romax} = k_{ro}(S_{wirr})$

maximum relative permeability to oil

nw

Corey coefficient for relative permeability to water

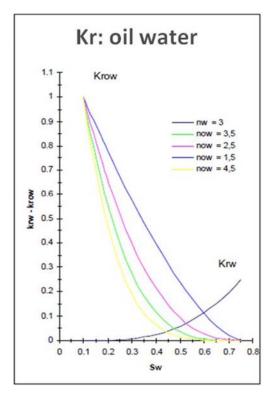
no

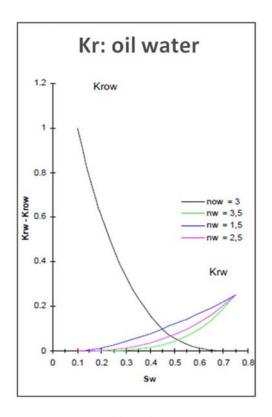
Corey coefficient for relative permeability to water

• Corey coefficients are always greater or equal to 1 and they control the shape of the relative permeability curves, hence they are related to the porous network (especially pore geometry) and the wettability

Relative permeability

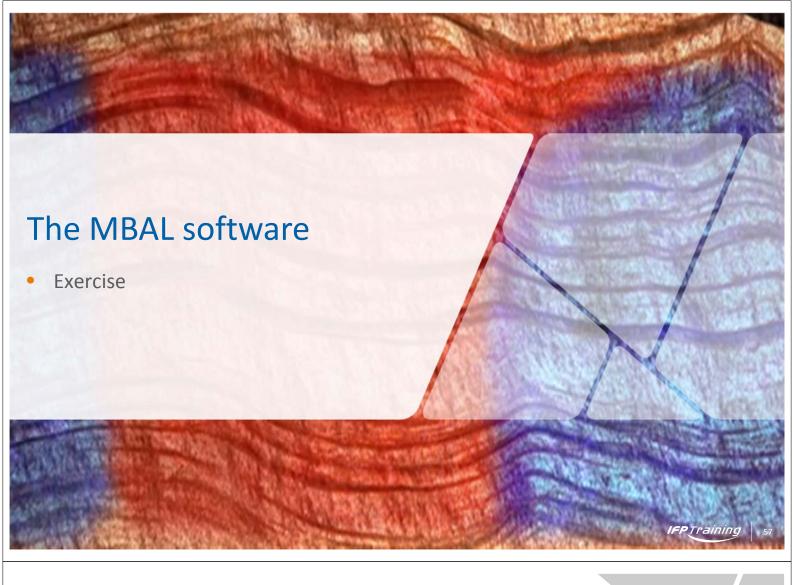
Corey normalization





Warning: the experimental relative permeabilities may differ from the Corey curves

Notes



Exercise – Reservoir 1

Data:

▶ Initial reservoir pressure: 5215 psig

► Reservoir temperature: 250 °F

Saturation pressure: 3600 psig

Oil gravity: 35 °API

GOR @ Psat: 800 scf/stb

Bo @ Psat: 1.456 rb/stb

Oil viscosity: 0.31 cP

Gas gravity: 0.78

Water salinity: 80000 ppm

Data:

► Reservoir porosity: 23%

▶ Connate water saturation: 15%

Estimated OOIP: 250 MMstb

► Aquifer activity: Radial aquifer, HVE modified model

Reservoir thickness: 100 ftReservoir radius: 2200 ft

• ro/ri: 5, θ : 180°, Kr_{aquifer}: 20 mD

► Relative permeabilities – end points:

	Residual saturation	End point	Corey exponent
Krw	0.15	0.6	1
Kro	0.15	0.8	1
Krg	0.02	0.9	1



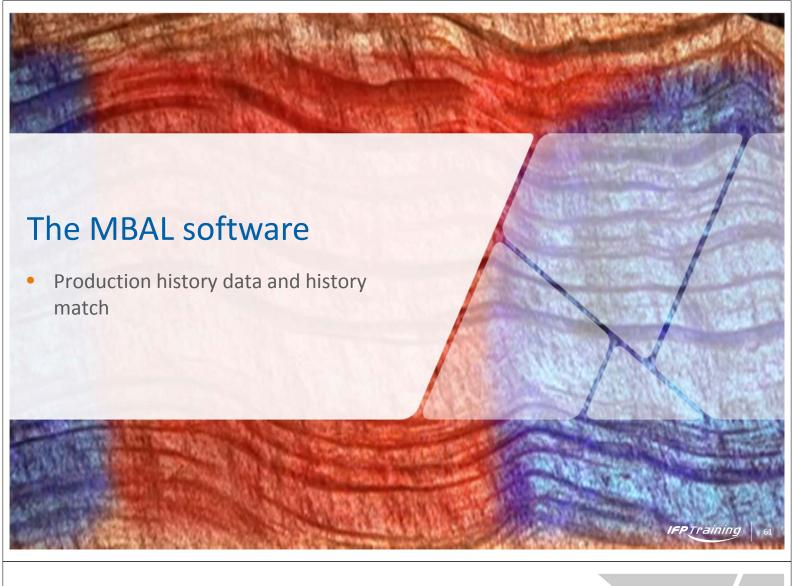
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Exercise - Reservoir 1

Production data:

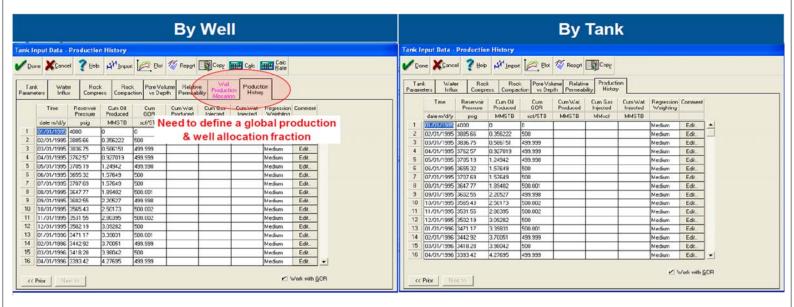
(see excel file: reservoir1.xls)

Date	Pressure	Cum Oil	Cum Gas	Cum Water
dd/mm/yy	psig	MMSTB	MMSCF	MMSTB
01/02/2000	5215	0	0	0
16/04/2000	5189,61	0,121533	97,2268	2,22E-05
15/06/2000	5176,5	0,218059	174,448	0,000102242
28/09/2000	5159,4	0,385867	308,693	0,000421904
12/12/2000	5149,07	0,504983	403,986	0,000803664
10/02/2001	5141,26	0,59986	479,888	0,0012039
11/04/2001	5133,65	0,694379	555,503	0,00168874
10/06/2001	5126,16	0,788544	630,835	0,00225786
09/08/2001	5118,73	0,882359	705,888	0,00291071
07/11/2001	5107,68	1,02243	817,946	0,00404538
21/01/2002	5098,53	1,13857	910,855	0,00513165
07/03/2002	5093,07	1,20799	966,395	0,00584408
06/04/2002	5089,43	1,25417	1003,34	0,0063441
21/04/2002	5093,7	1,25417	1003,34	0,0063441
06/05/2002	5097,08	1,25417	1003,34	0,0063441
21/05/2002	5099,89	1,25417	1003,34	0,0063441
20/06/2002	5092,9	1,30039	1040,31	0,00688304
03/09/2002	5079,75	1,41543	1132,34	0,00829563
17/12/2002	5065,04	1,57544	1260,36	0,0104555
17/03/2003	5053,71	1,71173	1369,39	0,0124862
01/05/2003	5048,22	1,77959	1423,67	0,0135643
14/08/2003	5035,65	1,9372	1549,76	0,0162417
27/12/2003	5003,17	2,21089	1768,71	0,021429
25/04/2004	4963	2,57733	2061,86	0,029275
24/07/2004	4939,32	2,84805	2278,44	0,0358439
22/10/2004	4917,52	3,11563	2492,5	0,0430237
04/02/2005	4893,09	3,42402	2739,22	0,0521613
05/04/2005	4879,39	3,59846	2878,77	0,0577414
20/05/2005	4869,2	3,72846	2982,76	0,0620932
01/07/2005	4859,76	3,84914	3079,31	0,0662827



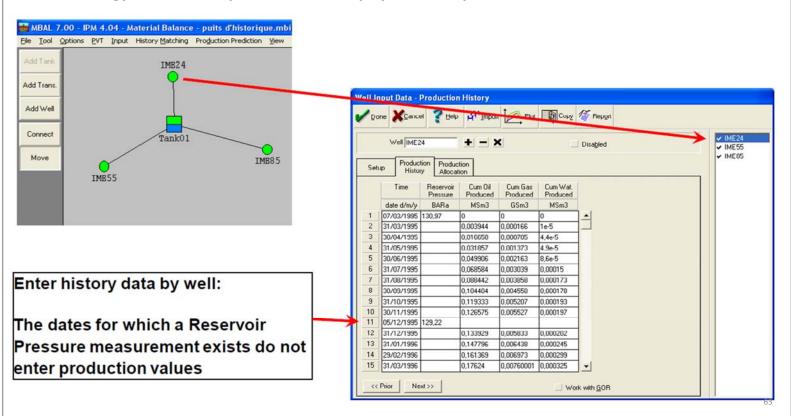
Input production history

- ▶ Depending on the 'Options' module choice, production/injection history can be introduced:
 - By well
 - By tank



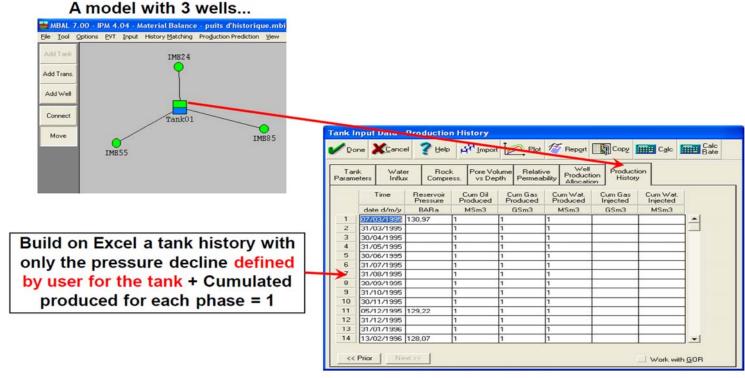
Input production history

► Methodology to introduce production history by well – step 1:



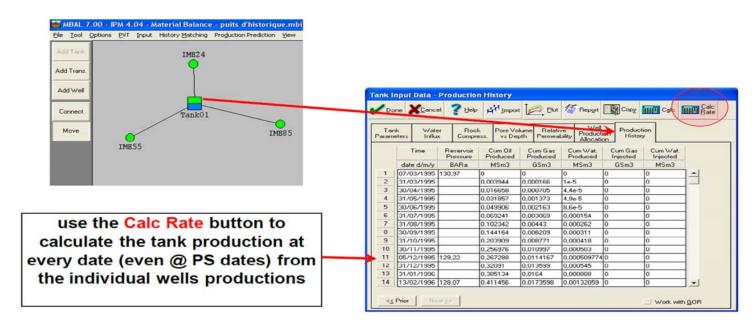
Input production history

▶ Methodology to introduce production history by well – step 2:



Input production history

▶ Methodology to introduce production history by well – step 3:





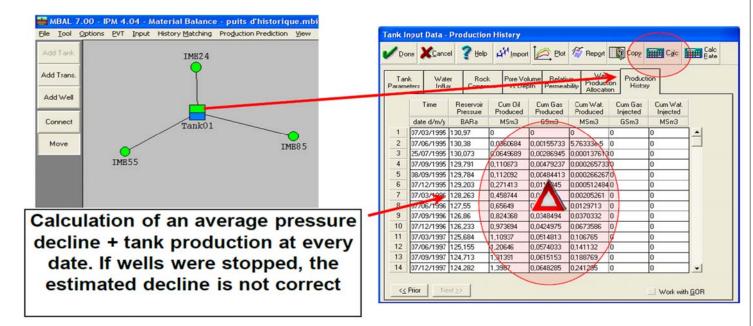
Tank pressure decline is unchanged: the values were previously defined by user

The 'Calc Rate' option re-calculates tank productions if the user changes the wells

production allocation → Recommended method

Input production history

- ▶ Conclusions on the methodology to introduce production history by well:
 - <u>Define "by hand" the average tank pressure decline, in Excel for example, and then use only</u> the 'Calc Rate' option for tank production generation.
 - Use of 'Calc' option not recommended...



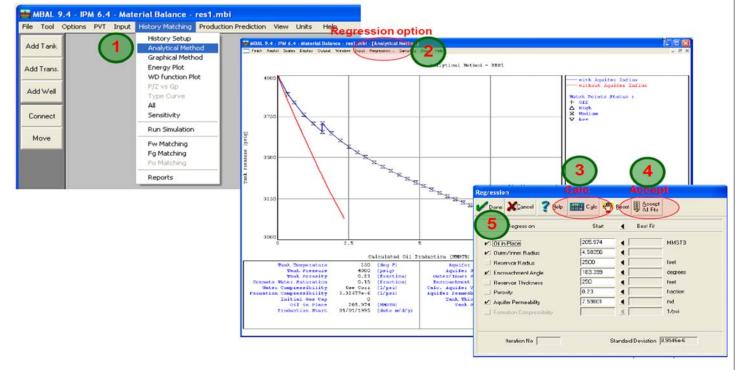
► Applications/Objectives:

- Find parameters to match pressure & production decline data by using
 - The analytical method
 - The graphical method
 - The fractional flow matching for predictions



6

History matching / Analytical method



- ▶ This method is a plot based method to assist on the estimation of unknown reservoir & aquifer parameters:
 - The plot shows the response of the model plotted against historical data
 - Calculated values: Oil production & aguifer water influx



- ▶ The graphical method plot is used to visually determine the different Reservoir and **Aquifer parameters**
 - Plots are based on the generalized linear Material Balance Equation:

$$F = N(E_o + mE_g + E_{f,w}) + W_e = NE_t + W_e$$

$$F = G(E_g + E_{f,w}) + W_e = GE_t + W_e$$

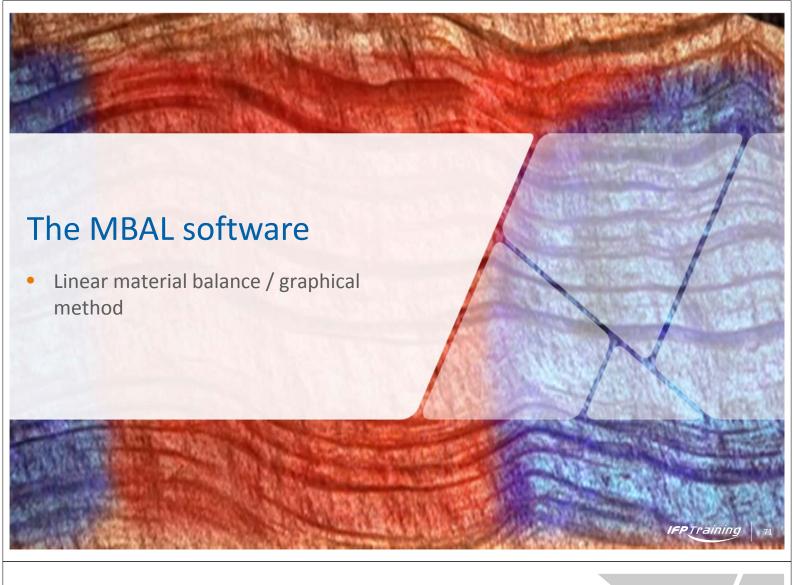


History matching/Analytical/Graphical methods

- ► Analytical / Graphical methods conclusion / recommendation:
 - These methods are more qualitative to determine the different reservoir and aquifer parameters.
 - To have a correct match we need both analytical & graphical methods fitted
 - Sometimes all the graphics are difficult to fit but they allow to understand the main physical parameters

The simulation step is only the final step of the matching process

IFPTraining



Linear Material Balance Equations

▶ In the years 1963-64, Havlena and Odeh proposed a new method to transform the previous global material balance equation in a linear equation:

$$F = N(E_o + mE_g + E_{f,w}) + (W_e)B_w$$

with

F Underground withdrawal

E Oil expansion & its original dissolved gas

 E_{g} Gas cap expansion

 $E_{f,w}$ Connate water expansion & pore volume reduction

N OOIP

W_e Aquifer water influx

$$N_p B_o$$

Produced gas (dissolved & injected gas)

$$(G_p - N_p R_s - G_i)B_g$$

Produced & injected water

$$(W_p - W_i)B_w$$



Linear material balance equations

- ► Fluids expansion:
 - Oil & dissolved gas expansion

$$E_o = (B_o - B_{oi}) + (R_{si} - R_s) \cdot B_g$$

Gas cap expansion

$$E_g = B_{oi} \left(\frac{B_g}{B_{gi}} - 1 \right)$$

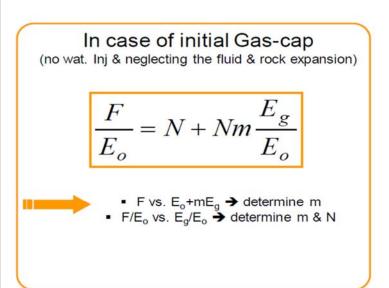
Rock & water expansion

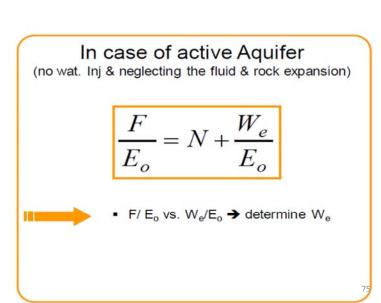
$$E_{f,w} = \left(1 + m\right) B_{oi} \left(\frac{C_w S_w + C_f}{1 - S_w}\right) \Delta P$$

Linear material balance equations & graphics

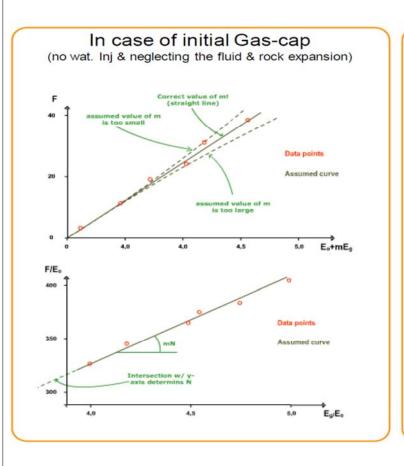
▶ Why this linear equation ?...

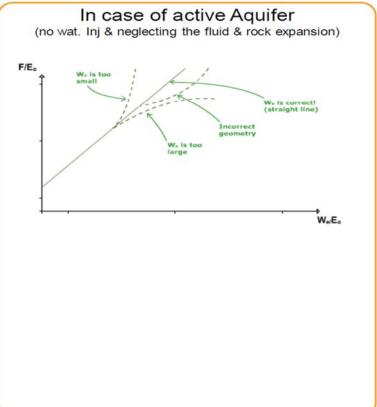
- This linear equation to have 'in case' a guick way to determine:
 - The initial Gas-cap size
 - The OOIP
 - The influence of water influx





Linear material balance equations & graphics





Linear material balance equations & graphics

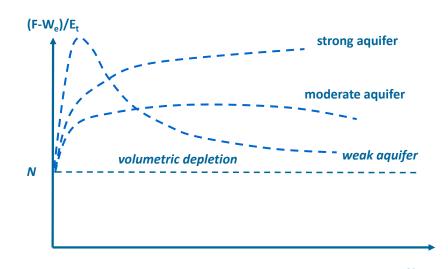
General case – Campbell plot

▶ In the most general case, MBE writes:

$$F = N(E_o + mE_g + E_{f,w}) + W_e \implies (F - W_e)/E_t = N$$

by plotting $(F - W_e)/E_t = f(N_p)$ we may have a straight line of slope 0 and intercept N In general, the plot is built without aquifer to check which type of aquifer should be added:

- strong aquifer => constant pressure boundary
- moderate aquifer => infinite aquifer
- weak aquifer => no flow boundary
- no aquifer => volumetric depletion



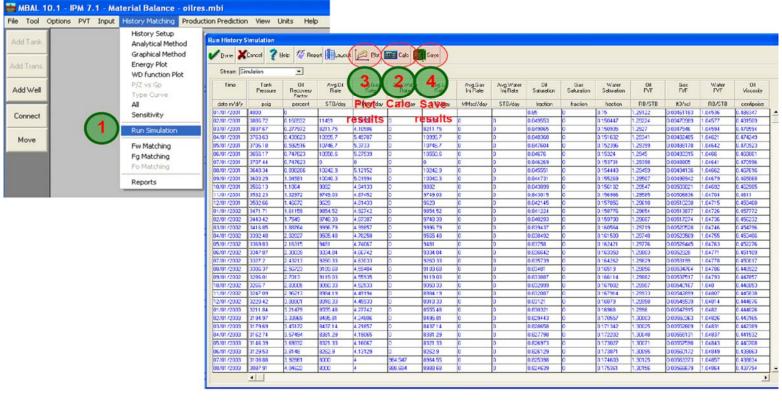
IFPTraining

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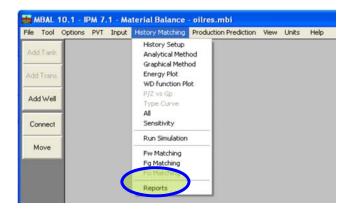
Notes



History matching / run simulation



History matching / Fractional flow



▶ One of the main difficulties when running a production prediction is to find a set of relative permeability curves that will result in a GOR, WCT or WGR similar to the history.

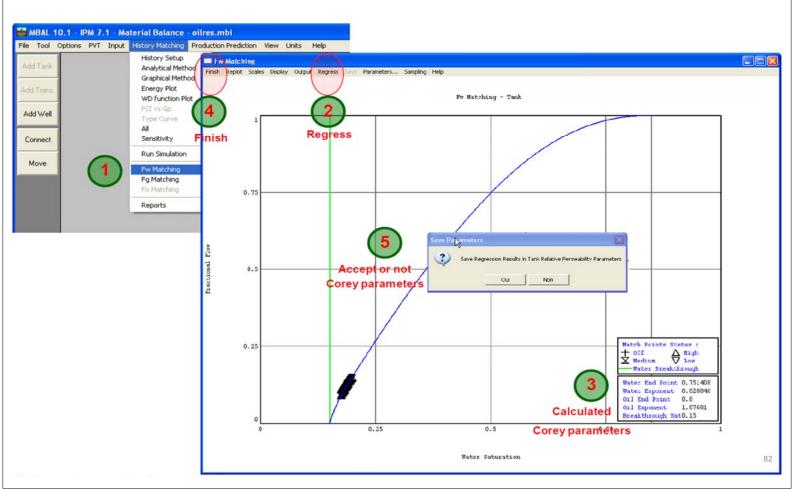
The purpose of this tool is to generate a set of Corey parameters that will reproduce the observed fractional flows

- → The fractional flow matching can only be used if a simulation has been run
- → It is also important to re-run a simulation each time an input data has changed



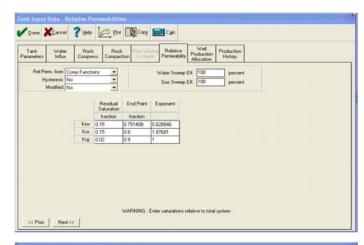
0.

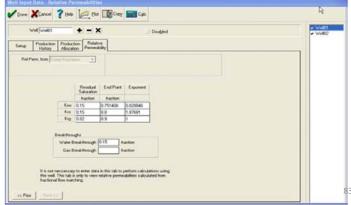
History matching / Fractional flow



History matching / Fractional flow

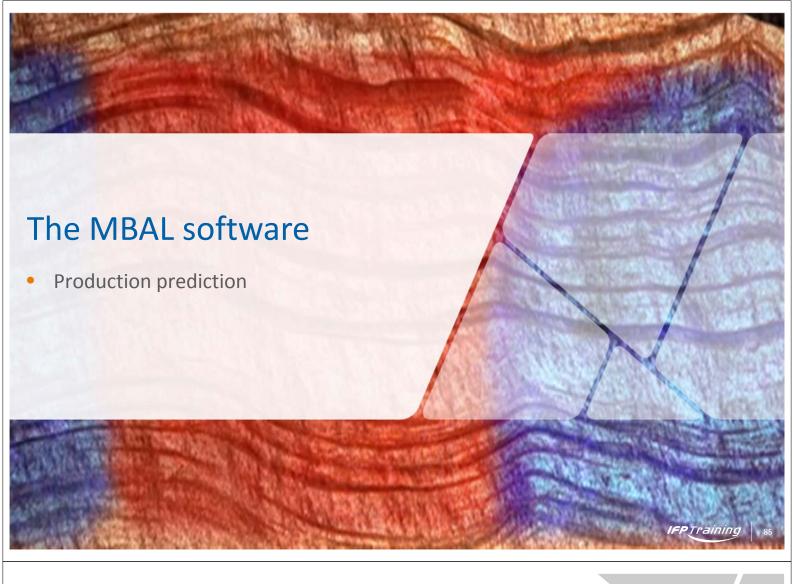
- ► In case of production history by tank (options menu)
 - Select the tank in which you want to make a Fw match and then save the Corey parameters in the corresponding tank
- ▶ In case of production history by well (options menu)
 - Select the tank in which you want to make a Fw match and then save the Corey parameters in the corresponding well



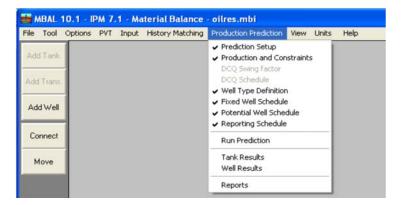


History matching – Methodology

- Use the analytical method
- ▶ Use the appropriate graphical method and determine if an aquifer must be introduced → Both analytical and graphical methods must be matched
- **▶** Test parameters impact:
 - first the aquifer parameters
 - then the OOIP or gas cap (interdependent parameters)
- ► Always check the matching on graphical methods
- Run the simulation
 - Control the match quality: comparison calculated/production pressure & other parameters (contacts, saturations...)
 - if same: good matching
 - if not: restart matching from analytical method (regression)



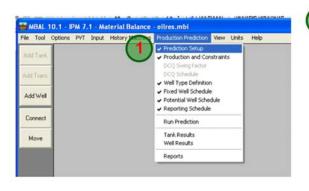
Production prediction

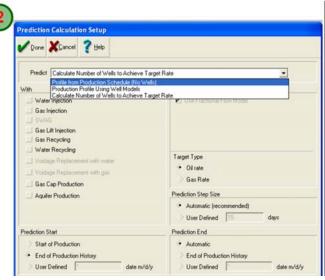


Applications/Objectives:

- Forecast the reservoir performance
 - After a history process to confirm remaining reserves
 - For pre-project field development studies
 - To test other reservoir production/injection strategies:
 - Start/stop gas re-injection
 - Increase/decrease water/gas injection
 - Start/stop existing/new wells

Production prediction / Prediction setup





▶ 3 possibilities to set up the prediction scenario:

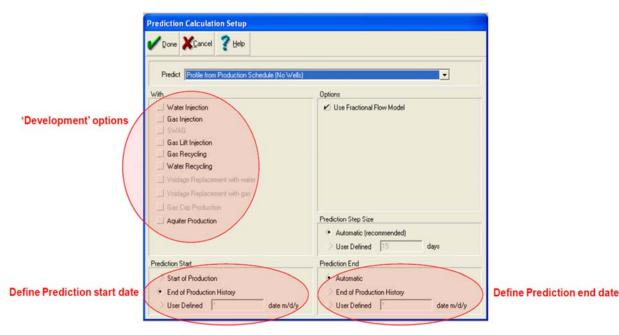
- Calculate production prediction without wells
 - prediction calculation @ tank (not available for multiple tanks model)
- Calculate production prediction using well models
 - prediction calculation @ well → need to define a well model with an IPR & a VLP model
- Calculate Number of wells to achieve a target rate
 - prediction to design a development scenario... → not used a lot...



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Prediction setup principle

The user can define manifold constraints independently to be consistent with the development scheme:



▶Depending on the 'development' options choice: all the relevant data will be entered in the "Production and Constraints" screen...



1st type: Profile from the production schedule (No wells)

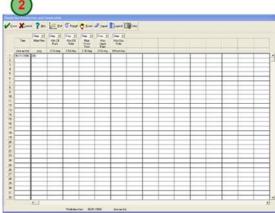
- Only for mono tank models.
- The well & manifold are completely ignored → No pressure losses
- Only the tank and the aquifer are taken into account.
- The user enters the tank production and injection schedule. The program simulates the tank and aquifer behaviors.
- Input data:
 - The tank parameters and relative permeabilities,
 - The aquifer type and parameters,
 - The description of the fluids injected (optional),
 - The production schedule for the main phase (e.g. oil for an oil system, gas for a gas or condensate system).
 - The injection schedule (optional)
- Assumptions:
 - The GOR, CGR, WC, WGR are calculated from the fractional flows using the tank relative permeabilities. These values then define the other phase rates (e.g. water rate for an oil system). Breakthroughs can also be entered to correct the tank relative permeabilities. There is no notion of abandonment.
- Calculated data:
 - The tank pressure and saturations,
 - Tank rates and cumulative productions for the other phases.
 - Tank average water salinity, gas cap gravity, etc.



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Production prediction – Production and constraints



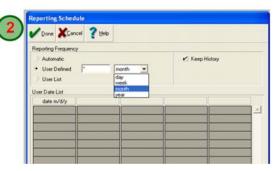


- ▶ The number and content of the columns vary depending on the prediction mode and Predict With options selected in the 'Prediction Setup' dialog box
- The following rules are applied:
 - The column is left entirely empty → There is no constraint on this parameter
 - A column contains only one value → This parameter will remain constant from that point onward

Production prediction – Reporting schedule

▶ Choose the way MBAL reports production prediction calculations





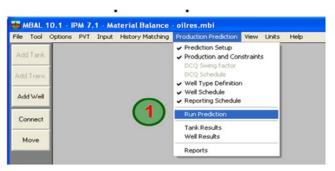
3 options for prediction reporting

- Automatic
 - The program displays a calculation every 90 days.
- User defined
 - Define any date increment in days, weeks, months or years.
- User list
 - A list of dates can be specified in the table provided. Any number of dates can be entered. The dates can be entered in any order MBAL will sort the dates into the correct order.
- •The option 'keep history' will keep history simulation within the production prediction calculation process



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Production prediction – Run prediction





Remark:

For now, with MBAL there is no restart option available... MBAL always re-does all the calculations during the complete prediction period...

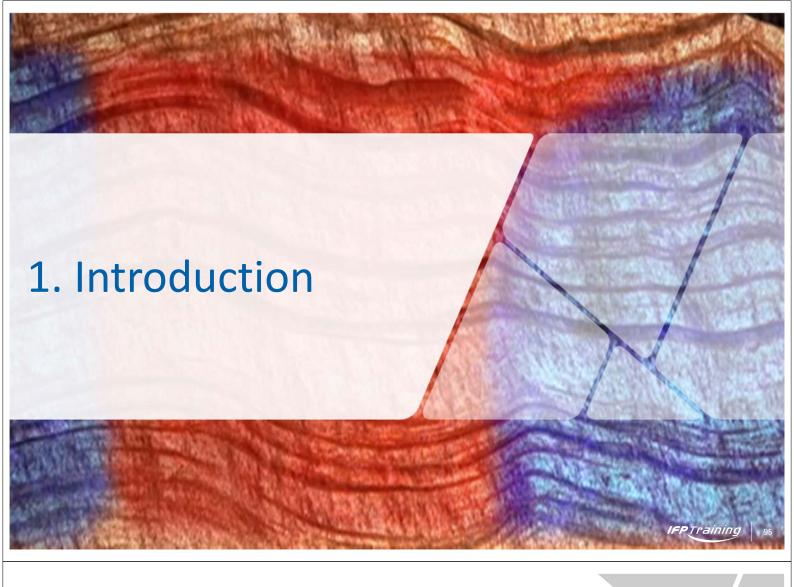


Enhanced Oil Recovery



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5. EOR projects implementation	215
6. Case study: Alwyn North – Brent East Panel	229



Production mechanisms

- Conventional methods
 - Natural energy
 - Water injection & immiscible gas injection

«Primary» «Secondary»

Unconventional methods

• E.O.R. Thermal processes

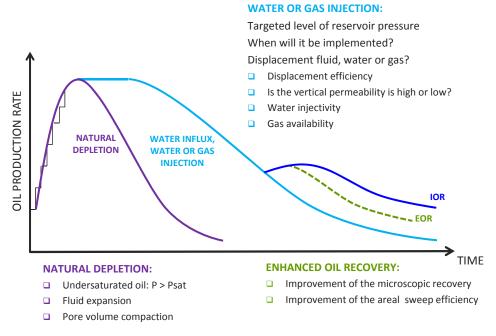
Chemical processes

Miscible gas injection

«Tertiary»

• Other technologies: Complex & intelligent wells

Improved reservoir management



IMPROVED OIL RECOVERY:

 Embraces all methods resulting in an increased recovery factor: infill drilling, high tech completion, artificial lift, treatment capacity...



Introduction

Sweep efficiency

▶ The sweep efficiency corresponds to the recovery factor (at reservoir conditions) for areas undergoing injection

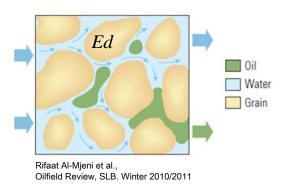
$$E = RF = \frac{Np.Bo}{Vp.So_i}$$

where So_i is the oil saturation at the start of injection

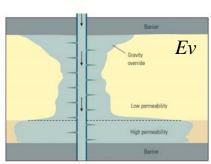
▶ The sweep efficiency can be expressed by: $E = E_d . E_a . E_v$

Ed is the displacement (or microscopic, Em) efficiency Ea is the areal efficiency

Ev is the vertical efficiency



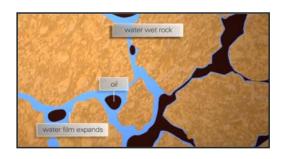


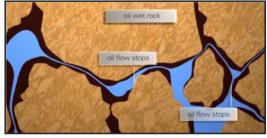


Displacement efficiency

- ▶ The displacement efficiency at the pore scale depends on:
 - The natural depletion
 - Wettability
 - The mobility ratio: relative permeability & viscosity

$$E_d = \frac{So_i - So}{So_i}$$





BP Videos – YouTube Exploiting science to increase oil recovery series

At the microscopic scale, oil can be trapped in the middle of pores when water flows around the oil in a water-wet formation. Oil that is connected to flow paths continues to be displaced.



0

Introduction

Displacement efficiency – Capillary number

► The capillary number (Nc) is a dimensionless ratio between the viscous forces and the capillary forces

$$N_c = \frac{v\mu}{\sigma}$$

The following formula is also valid:

$$N_c = \frac{k \left(\frac{\Delta p}{l}\right)}{\sigma}$$

Where:

 $\mu \to {
m Displacing}$ fluid viscosity

 $v \rightarrow \text{Darcy's velocity}$

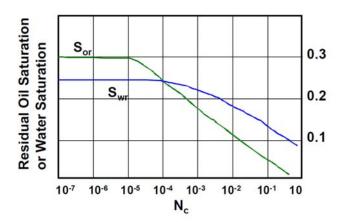
 σo Interfacial tension (IFT) between the displaced and the displacing fluids.

 $k \to \text{Effective permeability to the displaced fluid}$

 $\frac{\Delta p}{l} \rightarrow \text{Pressure gradient}$

Displacement efficiency – Capillary number

- ▶ A favorable capillary number can be achieved by:
 - Increasing the displacing fluid viscosity
 - Increasing the pressure gradient
 - Decreasing the IFT
- Nc to mobilize Sor is much higher than Nc at which it became trapped
- Nc vs Sor correlation varies by rock type
- ▶ Wetting phase residuals can be more difficult to mobilize.





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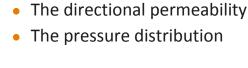
Introduction

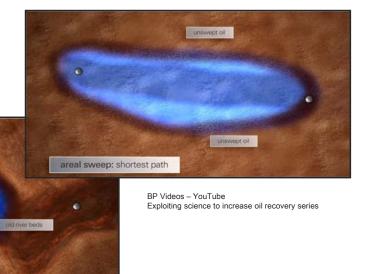
Areal efficiency

Areal efficiency depends on:

The pattern of injection

The mobility ratio: relative permeability & viscosity



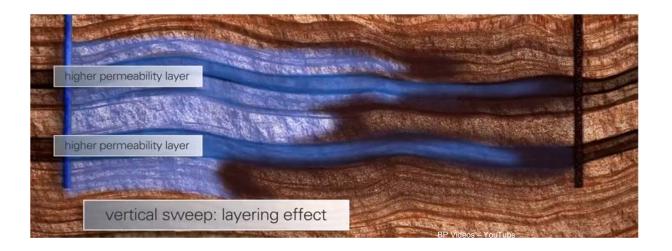


Oil can be bypassed because of inefficiencies in macroscopic sweep. A pattern flood can be affected by a heterogeneous formation or by fingering of a less viscous injectant into the oil.



Vertical efficiency

- ▶ The vertical efficiency depends on:
 - Rock properties variation between different flow units



The vertical sweep can be affected by viscous fingering, as well as by the preferential movement of the fluids along a high-permeability thief zone or by gravity override of injection gas or underride of injection water.



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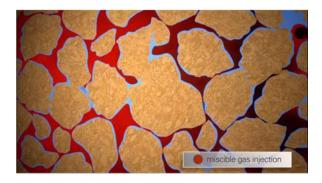
Introduction

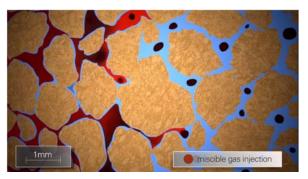
EOR principles

- ► Improvement of the displacement efficiency *Ed* by decreasing the residual oil saturation Sor (decrease in the interfacial tension)
 - Miscible gas injection
 - Surfactant injection chemical processes
- ▶ Improvement of the volumetric sweep efficiency Ea x Ev
 - ullet By increasing μw : polymer injection chemical processes
 - By reducing μo: thermal processes or miscible gas injection (CO₂)

Improvement of the displacement efficiency

▶ Residual oil saturation might be reduced through miscible gas injection, by reducing the interfacial tension the two phases (oil and gas) reach miscibility and Sor can be reduced until zero.





BP Videos – YouTube Exploiting science to increase oil recovery series

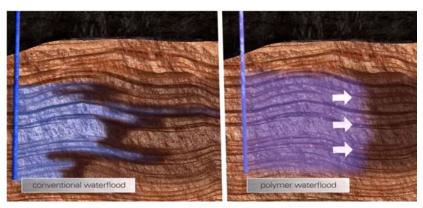


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Introduction

Improvement of the volumetric sweep efficiency

▶ When reservoir thickness is more important, the interfaces and the "fronts" can be unstable and subsequently distorted (tongue, fingering...)

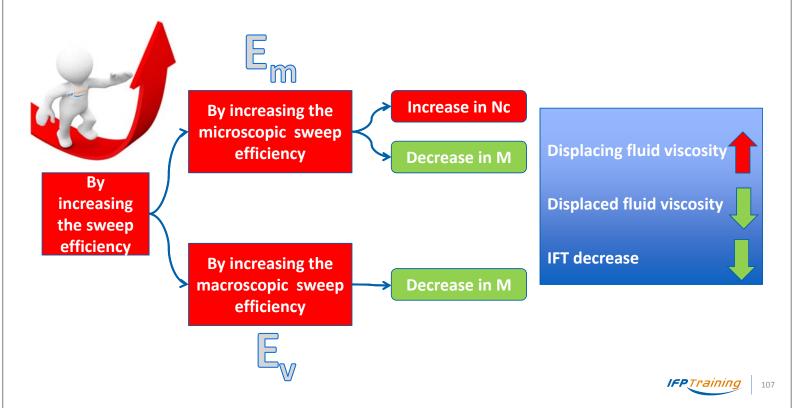


BP Videos – YouTube
Exploiting science to increase oil recovery series

▶ The stability of the displacement front is a function of the mobility ratio M

Improving the sweep efficiency – EOR methods

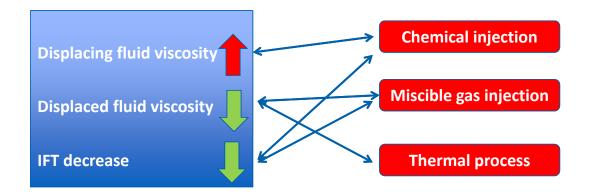
Objectives: How to increase the sweep efficiency (RF)?



Introduction

Improving the sweep efficiency – EOR methods

How do different EOR methods help increase the RF?



Pilots and studies

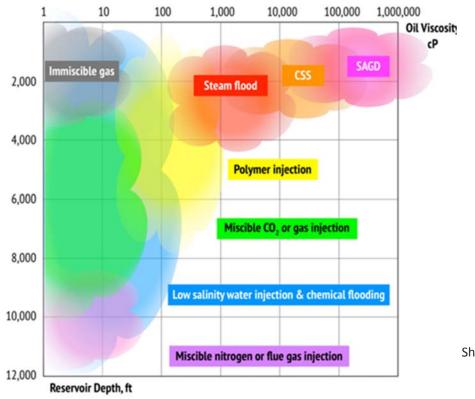
- Since EOR processes are often very expensive, economic studies are very important
- ▶ Pilot trials for some EOR processes are a must before going full field
- ▶ The project design should include detailed simulation studies (1D, 2D, 3D)
 - Numerical simulation of laboratory results
 - Mechanistic cross sections
 - Full field models
- Specific sophisticated experimental studies are needed
 - SCAL (wettability, Kr–Pc curves)
 - Waterflood and gasflood at reservoir conditions
 - Advanced PVT experiments to match EOS (e.g. swelling test)



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Introduction

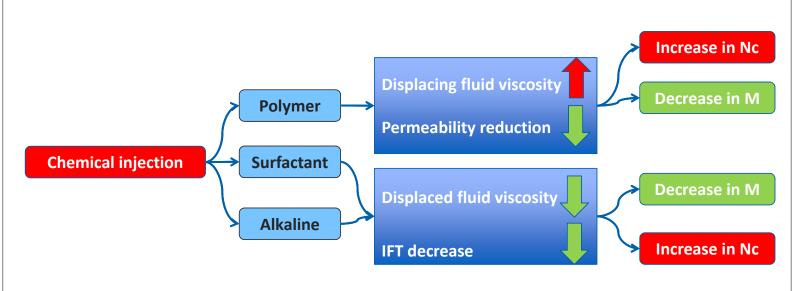
Scheme of screening for EOR methods



Shell, 2012



General impact of chemical flooding



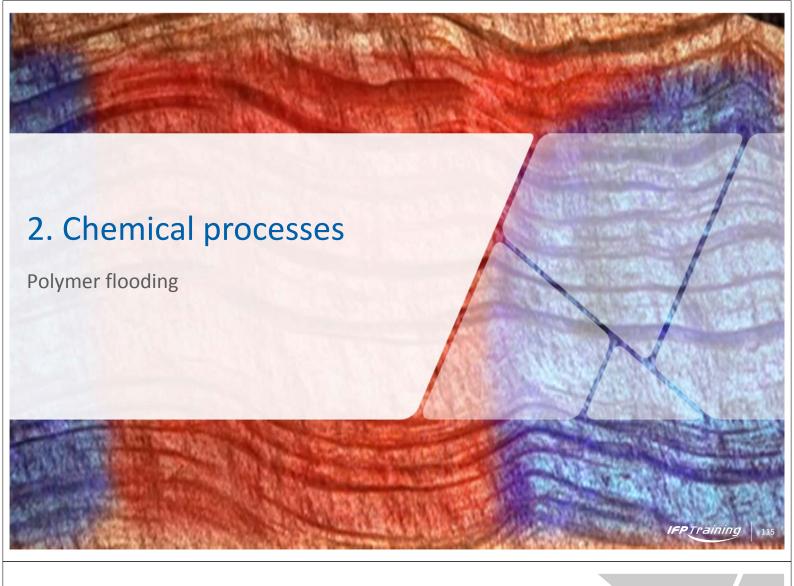
Polymers and surfactants

Chemical recovery methods have the following objectives:

- ▶ Polymers: to improve the volumetric sweep efficiency, by reducing the mobility ratio between the fluids injected and the fluids in place
- Surfactants: to eliminate or reduce the interfacial tension between oil and water and thus improve the displacement efficiency, i.e. maximize E_d
- ▶ To act on both phenomena simultaneously

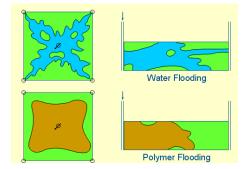


Notes



Injection of polymer solutions

- ▶ Polymer flooding is the most commonly used chemical enhancement process
- ▶ The displacing fluid is viscosified with soluble polymers, which reduces the mobility ratio and leads to a better volumetric sweep efficiency



- ▶ The recovery factor may be increased by a modest amount
- ▶ Polymer concentrations are between 100 to 1000 ppm and the treatment requires the injection of 15 to 30% PV followed by water injection

Polymer flooding

Resistance and permeability reduction

► Resistance factor (R_F)

• Ratio of brine mobility before polymer injection to that of a single-phase polymer solution flowing at the same conditions. It is an indication of the total mobility lowering contribution of a polymer.

$$R_{F} = \frac{\lambda_{w}}{\lambda_{p}} = \left(\frac{k_{w}}{k_{p}}\right) \left(\frac{\mu_{p}}{\mu_{w}}\right)$$

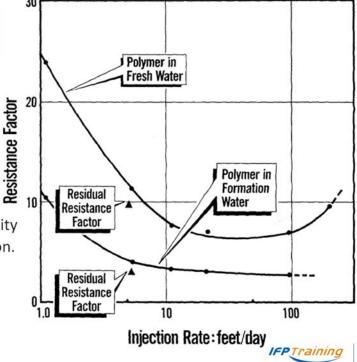
Permeability reduction factor

$$R_{k} = \frac{k_{w}}{k_{p}} = \left(\frac{\mu_{w}}{\mu_{p}}\right) R_{F}$$

Residual resistance factor

• indicates the permanence of the permeability reduction effect caused by the polymer solution. Brine mobility before polymer flood λ_{wb} Brine mobility after polymer flood λ_{wa}

$$\boldsymbol{R}_{RF} = \frac{\lambda_{wb}}{\lambda_{wa}}$$



Polymer flooding

Polymer stability

Different stabilities:

1. Mechanical strength

XG solutions are shear stable but not PAM. May be degraded in mixers, valves, pumps,

2. Thermal

XG and PAM are stable to above 200°F

3. Bacteriological

XG are sensitive and should be applied with Bactericides (Amine, Formaldehyde, ...)

4. Chemical

XG and PAM are compatible with a modest concentration of commonly dissolved ions in oilfield brines

PAM viscosity sensitive to pH

Mechanisms and technical screening

Increasing water viscosity (μ)

Possibly decreasing water permeability (k)

As a consequence decreasing water mobility (M_w)

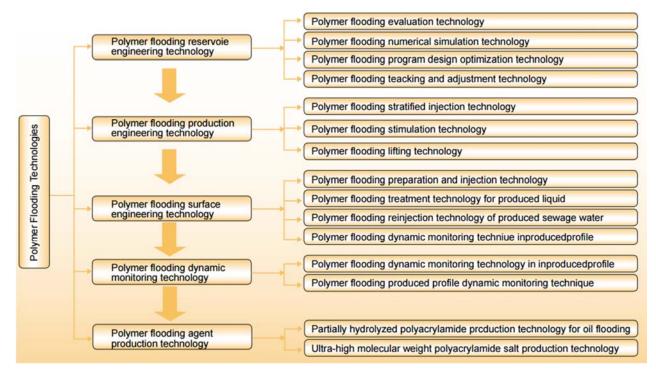
Parameters	(Kang, 2011)	(Goodlett, 1986)	(Taber, 1997)	(Al-Bahar, 2004)	(Alvarado, 2002)	Range field applie
Oil Viscosity (cP)	<200	<20	<150	<150	<100	1-80
Oil Gravity (°API)	-	>25	>15	-	>22	14-43
Oil Saturation (%)	-	>10	>50	>60	>50	50-92
Salinity (ppm)	<100,000	<100,000		<100,000	<100,000	
Hardness (ppm)	< 500	-	-	<1000	< 5000	
Wettability	12	water-wet preferred	-	-	220	
Depth (ft)	16	<9000	<9000	-	<9000	1300-9600
Formation Type	1.	sandstone preferred	sandstone preferred	-	sandstone preferred	
Temperature (°F)	<200	<200	<200	<158	<200	80-185
Permeability (md)	>10	>20	>10	>50	>50	10-15000
Porosity (%)	-	≥20	-	-	-	
Net Thickness (ft)	18	>10	-:	-	-	
Water Drive		-	7.1	no a	no a	
GOR	- 6	-	-	<10	-	



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Polymer flooding

Technology



▶ Limitations

- High oil viscosities require a high polymer concentration
- Results are normally better if the polymer flood is started before WOR becomes excessively high
- Clays increase polymer adsorption
- Some heterogeneity is acceptable

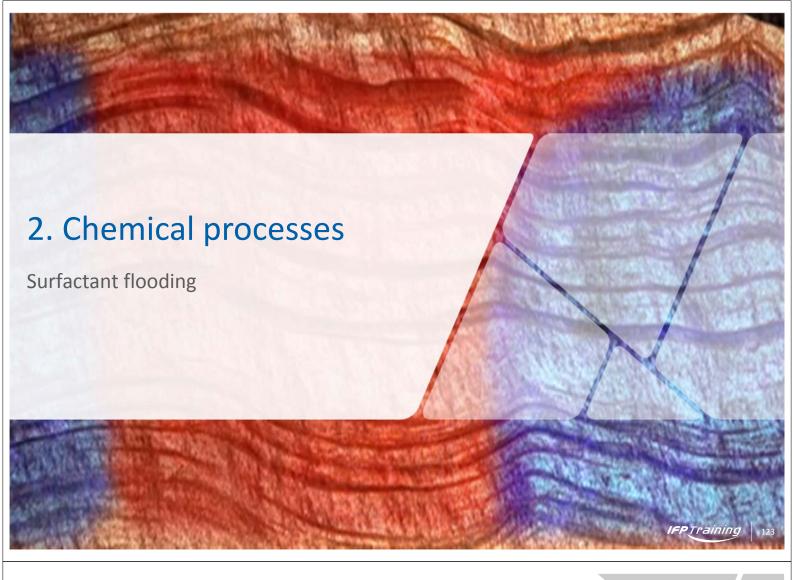
▶ Challenges

- Lower injectivity than water injectivity
- The stability of synthetic polymers is lower in case of shear degradation, salinity and divalent ions.
- Biopolymers cost more and are subjected to microbial degradation and have greater potential for wellbore plugging



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Notes



Surfactant flooding

Surfactant performance is optimal under a narrow range of conditions

- Difficult at high temperature and high salinity
- Preferably sandstones (some surfactant like alkali may cause precipitation in carbonates reservoir resulting in pore plugging)
- Surfactants may have low viscosity leading to a poor sweeping efficiency

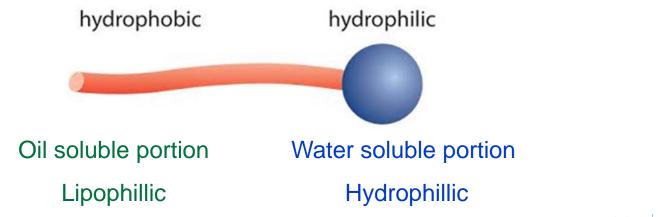
▶ However, surfactant flooding has a high potential in terms of oil recovery

- Surfactant flooding is generally used with the injection of other chemicals that reduce surfactant losses due to the adsorption on the reservoir rock
- The latest technology is a combination of Alkaline, Surfactant and Polymer (ASP flood): nowadays, the ASP is the recommended process

Surfactant flooding

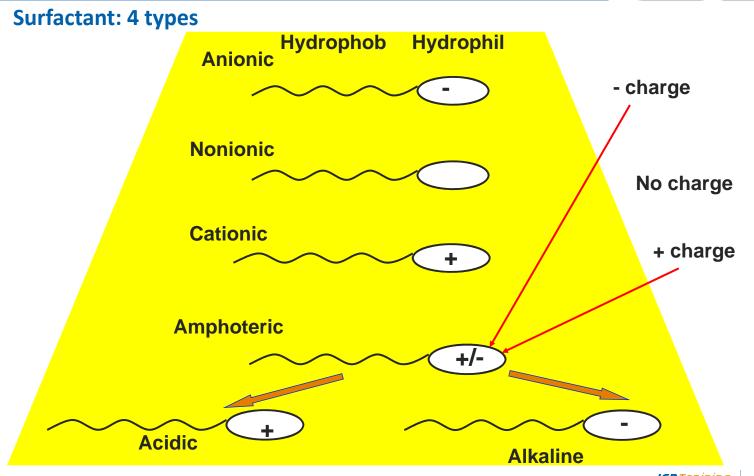
What is a surfactant?

- ▶ It is a "surface active agent" (very old surfactant is SOAP)
- ▶ A chemical compound that combines oil soluble and water soluble properties
- Surfactants are "active" at a surface or interface



IFPTraining 125

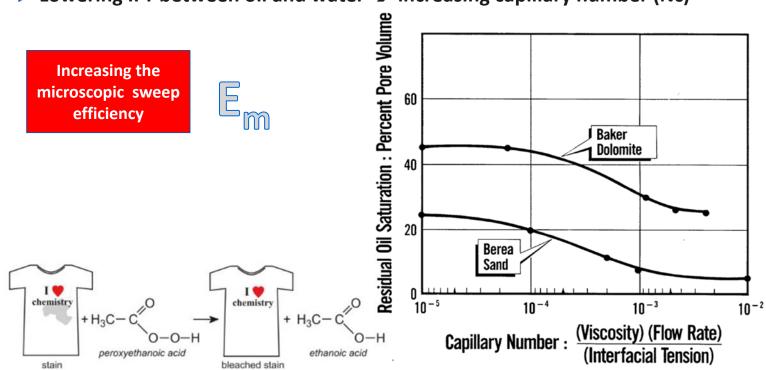
Surfactant Flooding



Surfactant Flooding

Mechanisms

▶ Lowering IFT between oil and water → increasing capillary number (Nc)



(Burnett and Dann, 1981)



Critical micelle concentration (CMC)

(W)

(W)

(W)

(W)

(W)

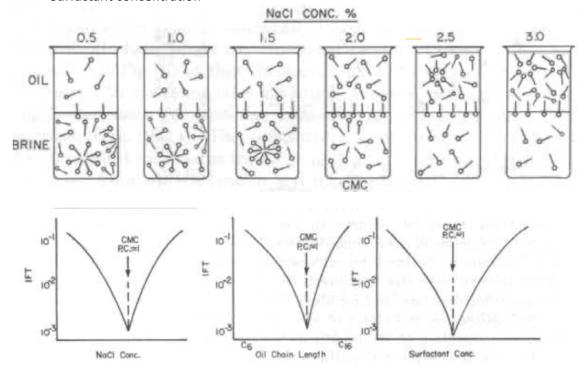
(W)

(CMC - critical micelle concentration

Critical micelle concentration (CMC)

▶ Parameters affecting CMC

- Salinity
- Oil molecule length
- Surfactant concentration



IFPTraining

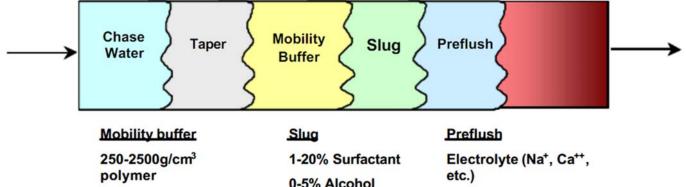
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Surfactant Flooding

▶ Range of oil field characteristics with microemulsion flooding

Property	Range		
Depth [ft]	350 - 4550 (107 - 1387 m)		
Reservoir Temperature [°F]	55 - 200 (12.8 - 93 °C)		
Porosity [%]	13 - 32		
Permeability [md]	7 - 300+ {avg.}		
Type of Reservoir	Unconsolidated to well cemented sandstones, limestones		
Formation Water [ppm TDS]	3000 - 160,000 (3000 - 160,000 mg/kg)		
Hardness [ppm Ca, Mg, Fe]	25 - 5000 (25 - 5000 mg/kg)		
Crude Gravity [°API at 60°F]	15 - 45 (0.965 - 0.801 g/cm ³)		
Crude Viscosity [cp]	3 - 31.7 (3 - 31.7 m Pa*s)		
Crude Type	Aromatic-Paraffinic-Naphtenic		

▶ Micellar polymer injection process



Stabilizers

0-1% Alcohol

Biocide

0-100% V_{pf}

1-20% Surfactant 0-5% Alcohol 0-5% Cosurfactant 0-90% Oil

Polymer 5-20% V_{pf} Sacrificial chemicals

0-100% V_{pf}



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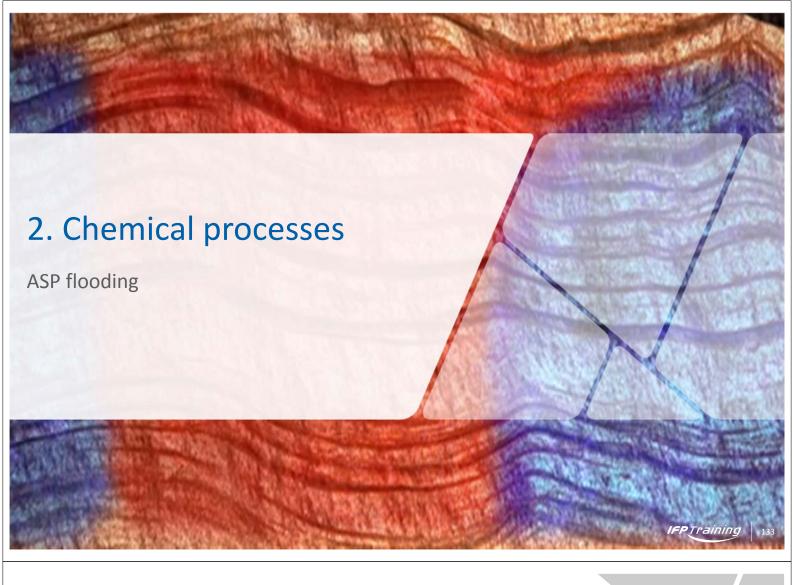
Surfactant flooding

Limitations

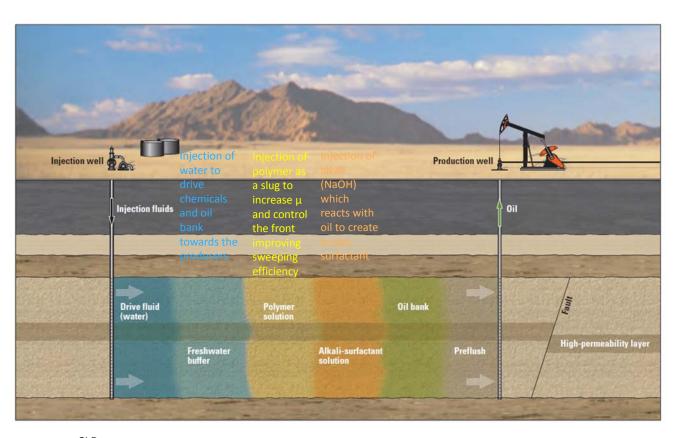
- Low-moderate salinity
- Moderate temperature
- Clean sandstone
- No anhydrite
- Water wet
- Med-high permeability
- Homogeneous
- Onshore

▶ Challenges

- High salinity
- Low or high temperature
- Carbonate
- Anhydrite
- Oil wet
- Low permeability
- Fractured
- Offshore
- Do research



ASP flooding



Chemical processes

ASP flooding

Principles: combining the best techniques

- Injection of alkali (typically sodium hydroxide) which reacts with acidic oil components to create in-situ surfactant (petroleum soap)
- Simultaneous injection of synthetic surfactant to reduce IFT
- Injection of a water-soluble polymer both with the alkali-surfactant mixture and as a slug, following the injection of chemicals in order to increase viscosity and control the flooding front, thus improving sweeping efficiency (mobility buffer)
- Injection of water to drive chemicals and oil bank towards the producers

Performances

- ASP can theoretically lead to very high recovery factor, up to 90% as shown in laboratory and field pilot
- ASP is not recommended for carbonate reservoirs (possible reaction of alkali with calcium ions to form precipitates)



Alkali Flooding

Mechanisms

- ► Lowering IFT between oil and water → increasing capillary number (N_c)
- ▶ The difference between micellar flooding and alkaline is that in micellar flooding the surfactant is injected, while in alkaline (or caustic) flooding the surfactant is generated in situ
- ▶ A high pH chemical EOR method



NaOH

Alkaline Flooding

How does alkaline generate a surfactant?

- Alkaline method needs an oil with acidic nature
- No acidic species in oil → No surfactant can be generated
- Oil characteristic determination is essential to apply the alkaline method
- ▶ The attractiveness of an oil for alkaline flooding is given by its acid number. The acid number is the milligrams of potassium hydroxide (KOH) needed to neutralize one gram of crude oil.
- Of course salinity and temperature are also effecting factors



Micellar/polymer, ASP and Alkaline Flooding

Crude Oil Condition

- ▶ Gravity: °API
- Viscosity: cp
- Composition

Reservoir condition

- Oil saturation: %
- Type of formation
- Net thickness: ft
- Average permeability: md
- Depth: ft
- Temperature: °F

- > >20
- <35
- Light intermediate are micellar/polymer. Organic acids needed achieve lower IFT with alkaline methods
- >35
- Sandstones preferred
- Not critical
- > >10
- <9000
- **<200**

Key points to keep in mind



Chemical processes

Chemical process objectives may be quite different

- Polymer flooding: increases viscosity of the displacing fluid in order to stabilize the front and increase the volumetric efficiency
- Surfactant flooding: decreases IFT in order to decrease the residual oil saturation and increase the microscopic efficiency
- Alkali-Surfactant-Polymer flooding combines both Polymer flooding and Surfactant flooding to reach a very high RF, up to 90%

Operational conditions

- The proper design of the process may be difficult
- Temperature and salinity may severely decrease the process efficiency
- Alkali and ASP flooding shall not to be used with carbonates reservoir as they may cause precipitation and pore plugging



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Key points to keep in mind



Polymer flooding

Principles

• Displacing fluid is viscosified with soluble polymers, which reduces the mobility ratio and leads to a better volumetric sweep efficiency

Limitations

- High oil viscosities require a high polymer concentration
- Results are normally better if the polymer flood is started before WOR becomes excessively high
- Clays increase polymer adsorption
- Some heterogeneity is acceptable

Key points to keep in mind



Surfactant flooding

Principles

• Surfactants: to eliminate or reduce the interfacial tension between oil and water and thus improve the displacement efficiency, i.e. maximize E_d

Limitations

- Difficult at high temperature and high salinity
- Preferably sandstones (some surfactant like alkali may cause precipitation in carbonates reservoir resulting in pores plugging)
- Surfactants may have low viscosity leading to poor sweeping efficiency



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Key points to keep in mind



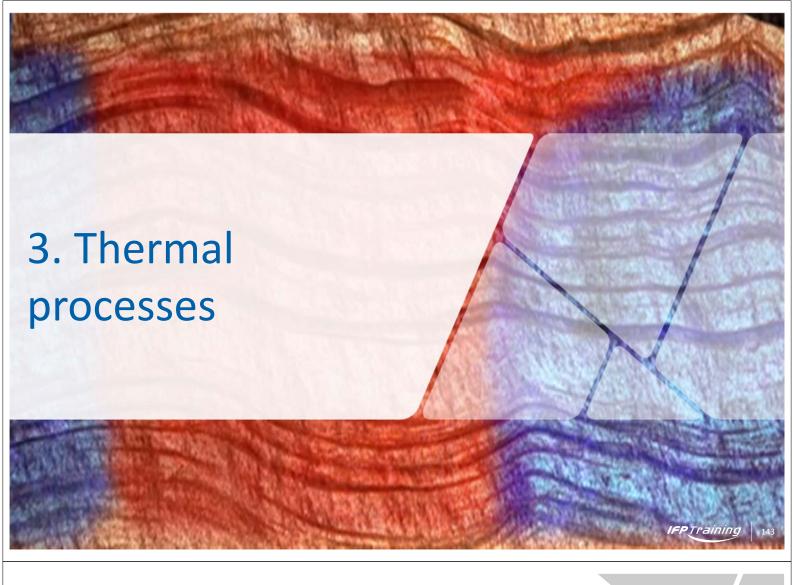
Alkaly-Surfactant-Polymer flooding

Principles

- Injection of alkali (typically sodium hydroxide) which reacts with acidic oil components to create in-situ surfactant (petroleum soap)
- Simultaneous injection of a synthetic surfactant to reduce IFT
- Injection of a water-soluble polymer both with the alkali-surfactant mixture and as a slug following the chemicals injection, in order to increase viscosity and control the flooding front, thus improving sweeping efficiency (mobility buffer)
- Injection of water to drive chemicals and oil bank towards the producers

Limitations

 ASP is not recommended for carbonate reservoirs (possible reaction of alkali with calcium ions to form precipitates)



Thermal methods

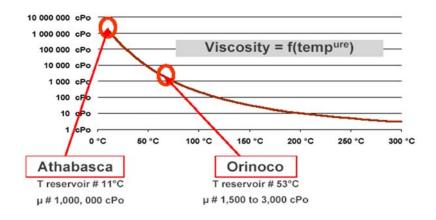
Objectives

- To get a lower oil viscosity → higher mobility
- To improve well productivity
- To improve the final recovery factor

Different Methods

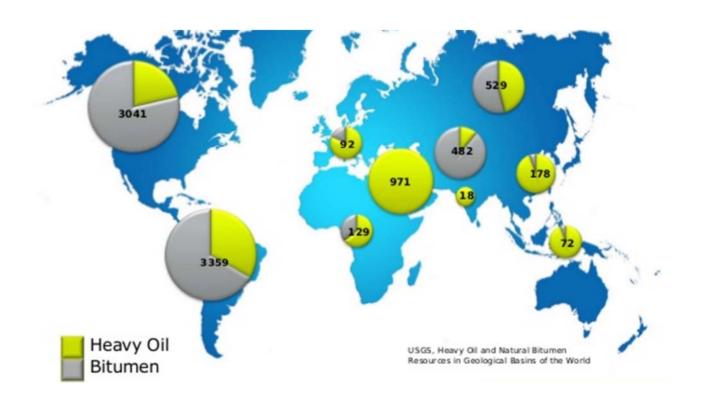
- Steam flood
- In situ combustion

▶ These methods are used with viscous oil



Thermal methods

Heavy oil and bitumen deposits (Billion BBL)





1

Thermal methods

Principals of heat transfer

$$q = k_h A \frac{dT}{dx}$$

q = rate of heat transfer in the x direction, BTU/hr

kh = thermal conductivity, BTU/hr-ft-°F

A =area normal to x direction, ft^2

T = temperature, °F

x = length, ft

Convection

$$q = hA(Tf - Ts)$$

h = film heat transfer coefficient, BTU/hr-ft²-°F

A = area of heat transfer surface, ft²

Ts = temperature of solid, °F

Tf = temperature of fluid, °F

Thermal methods

Drive mechanisms

- ▶ Oil viscosity reduction
- ▶ Thermal expansion
- Distillation
- Solution gas drive
- Emulsion drive



Thermal methods

Steam flood

- Steam generated at the surface is injected into the reservoir through specially distributed injection wells.
- Different ways of implementation:
 - Cyclic steam injection (Huff and Puff),
 - · Continuous steam flood,
 - Steam assisted gravity drainage

Steam injection

Several methods

Cyclic steam injection (huff and puff)

- Injection of steam into one well (10 days 1 month)
- Soaking period: well shut-in (1 10 days)
- Production from the stimulated well (3 months 1 year)
- Stimulates production, accelerates depletion
- Not a recovery technique except in specific cases

Steam drive

- Continuous injection of steam into injectors
- Oil pushed to producers
- Recovery technique
- Often applied after depletion by several "huff and puff" cycles

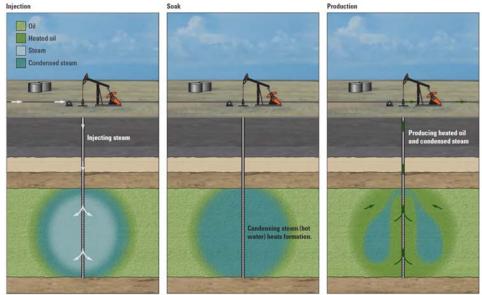
SAGD process

- Steam assisted gravity drainage
- Continuous injection of steam into horizontal wells
- Oil produced by gravity in horizontal wells located below the injectors



Cyclic steam stimulation

- ▶ High pressure steam injected during several weeks --> heating of the oil, reduction of viscosity
- Soak period during several weeks
- Pumping of the oil up to the surface
- When production declines: switch back to injection



▶ Proven technology

Examples:

• Canada: Cold Lake, Wolf Lake, Primrose

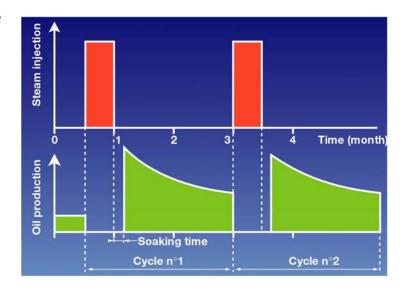
• Venezuela: Maracaibo area

• California: Kern River

Operating costs: 4-5 US\$/bbl

Drawbacks:

- Only stimulation around the wellbore
- Limited recovery factor (15-20%)
- Energy consumption and GHG emission

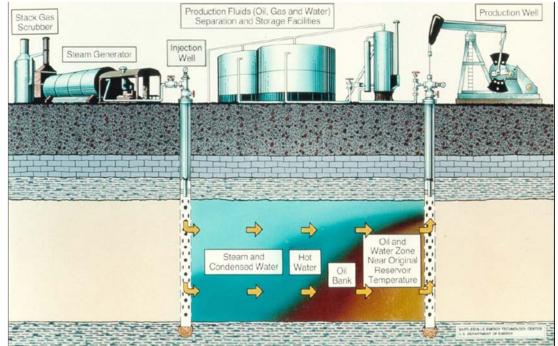




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Continuous steamflood

- ▶ High-temperature steam is continuously injected into the reservoir
- > As the steam looses heat to the formation, it condenses into hot water
- ▶ Steam and hot water drive to move the oil to production wells

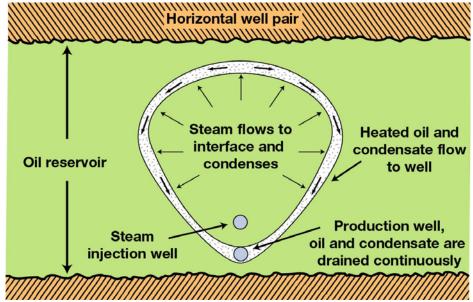


- As the formation heats, oil recovery is increased by:
 - The viscosity reduction, increasing oil mobility
 - The expansion or swelling of the oil
 - The vaporization of lighter fractions of the oil. The fractions move ahead into the cooler formation where they condense and form a solvent or miscible bank
 - Condensed water forms a waterflood
- ▶ Up to 50% recovery can be achieved with a oil/steam ratio (OSR) of 0.2
- Examples: Maracaibo (Venezuela), California (Kern River), Indonesia (Duri), **Alberta (Peace River)**
- Often used after initial CSS phase (to stimulate well neighbourhoods)



Steam Assisted Gravity Drainage

▶ In the SAGD process, two parallel horizontal oil wells are drilled in the formation, one about 4 to 6 m above the other. The upper well injects steam, possibly mixed with solvents, and the lower one collects the heated crude oil that flows out of the formation, along with water from the condensation of injected steam



Steam injection

Heat losses

- Heat is carried over some distance by the displacing fluid to its final destination in the reservoir
- Heat loss is a critical factor for recovery processes by hot fluid injection
 - Heat loss from the reservoir to the surrounding formations: the extent of the steam condensation zone is reduced and so is the thermal efficiency of the process
 - The consequence is that it is not applicable to very thin beds or to reservoirs with a large spacing between the injector and the producer
 - Heat loss from the well: a further cause of heat loss occurs in the passage of the hot fluids in the injection well from surface to the injection zone.
- It should be mentioned that steam generation is intensive in terms of
 - Energy consumption and combustion of hydrocarbons
 - Environmental impact due to CO₂ produced in the above combustion
 - Use of fresh water, which can be scarce, treatment and re-cycling of produced water
- ▶ The rewards are that the steam injection can yield high recovery factors

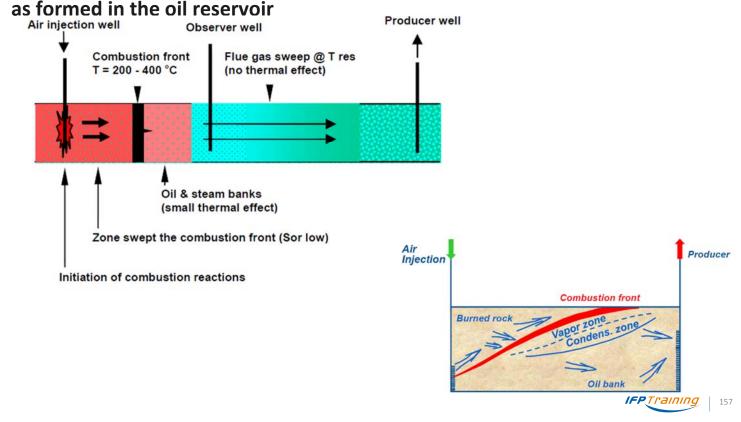


In-situ combustion

- Oil is ignited around well bore
- Burning front sustained by continuous injection of air
- ▶ A small portion of the oil is burned
- The heat generated
 - Reduces oil viscosity
 - Produces miscible fluids
 - Increases sweep efficiency
 - Reduces oil saturation
- Continuous air injection develops efficient gas drive mechanisms

In-situ combustion: air injection

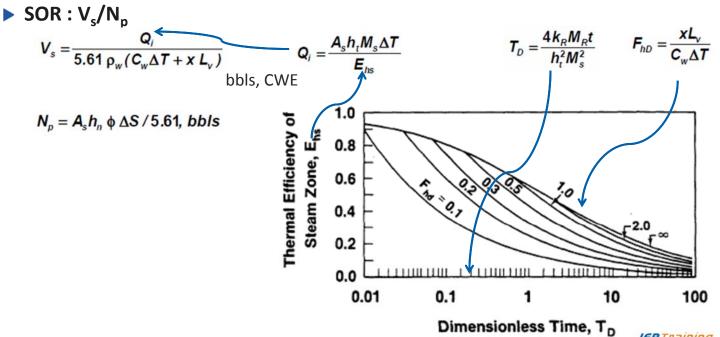
▶ Schematic representation of in-situ combustion process and the various zones



Continuous steam injection

Performance prediction

- Myhill and Stegemeier's Method
- The performance is presented by steam oil ratio SOR (normally between 2 to 8)



- h_n = net sand thickness [ft]
- f = average porosity of sand [fraction]
- ΔS = average change in oil saturation during steamflood [fraction]
- x = downhole steam quality
- L_v = latent heat of vaporization at downhole conditions [BTU/lbm]
- h_t = gross sand thickness [ft]
- M_s = average heat capacity of steam zone [BTU/ft3-°F]
- $\Delta T = T_S T_R$ [°F]
- T_S = steam zone temperature [°F]
- T_R = original formation temperature [°F]
- E_{hs} = thermal efficiency of steam zone
- M_R = average heat capacity of cap and base rock [BTU/ft3-°F]
- t = time of steam injection [hr]
- k_R = thermal conductivity of cap and base rock [BTU/ft-hr-°F]
- Cw: Specific heat of water over the temperature range corresponding to ΔT

Enhanced Oil Recovery - Thermal Method



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Key points to keep in mind



Thermal processes

- ► The main objective of thermal processes is to decrease oil viscosity in order to increase oil mobility
- Mandatory in some cases, especially with heavy oils that may not flood in local normal conditions (typically in Canada)
- **▶** Drive mechanisms:
 - Oil viscosity reduction
 - Thermal Expansion
 - Distillation
 - Solution gas drive
 - Emulsion drive



Thermal processes

- Several processes may be used
 - Cyclic steam injection
 - Steam flooding
 - SAGD
 - In-situ combustion
- ► Main drawbacks of thermal processes: economics
 - Heat losses
 - Energy to produce steam
 - Water: treatment, recycling



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Notes



Generalities

Definition

- Two fluids are miscible if they can mix in all proportions and form a single homogeneous phase
- The minimum miscibility pressure is the lowest pressure at which miscibility (direct or multiple contact) can be achieved, at given temperature and composition

Miscible gas injection

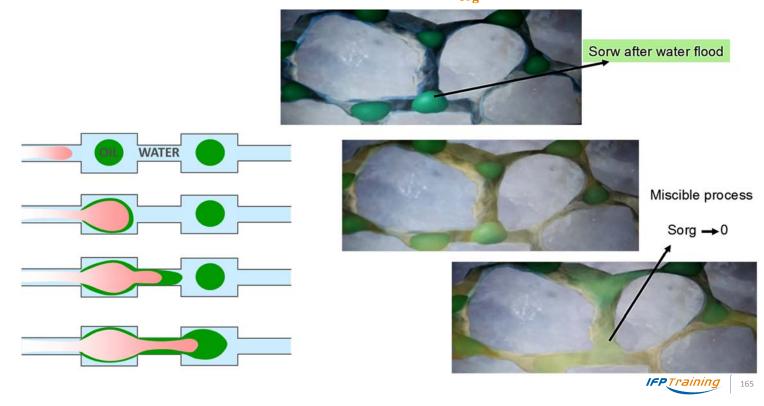
- No more interfacial tension: S_{org} tends to zero
- Direct (first contact) miscibility: rare
- Multiple contact (dynamic) miscibility
 - Vaporizing gas drive
 - Condensing gas drive

Water Alternating Gas

To improve miscible gas flooding stability

Principles of the miscible gas injection

- Pore-level mechanism: microscopic efficiency
 - Mobilization of oil that was trapped behind water front: oil swelling
 - Formation of gas-oil interfaces \rightarrow reduction of IFT \rightarrow S_{org} tends to zero



Miscible gas injection

Gas-Oil miscibility

- ▶ The miscibility depends on pressure and temperature
- ► The miscibility is rarely obtained directly: multiple-contact miscibility (also called dynamic miscibility)
- Multiple-contact miscibility: the injected gas and the in-situ oil exchange components until miscibility between the two phases is reached
- **▶** Two-types of multiple-contact miscibility:
 - Vaporizing gas miscibility
 - Condensing gas miscibility

 (At reservoir temperature)

 1 First contact miscibility
 2 Vaporizing gas drive miscibility
 3 Condensing gas drive miscibility
 4 Immiscible fluids

Pressure (Log. P)

ADVANTAGES

Good microscopic recovery:

- Low residual oil Saturation
- Low interfacial tension

▶ Good volumetric efficiency if:

- Gravity stable displacement
- Miscible displacement
- Mobility control (WAG)

Phase behavior

- Low viscosity
- High relative permeability
- Then high injectivity

Thermodynamic exchanges

- Oil swelling
- First contact miscibility
- Multi contact miscibility

DRAWBACKS

- Reservoir Heterogeneity
 - High sensitivity to gas sweep

Unfavorable mobility ratio

- Unstable displacement
- Poor sweep efficiency
- High compression cost
- Gas availability



Gas Injection: first screening

Nature of gases:

- Hydrocarbon (Lean, Rich, Enriched)
- Non Hydrocarbon: CO₂, N₂, Air, Flue Gas

▶ Nature of the gas / rock / fluids reactions:

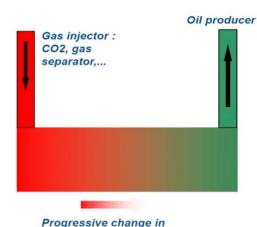
- Exchanges (mass transfer) = Important or not
- Thermal effects or not (O₂ presence)

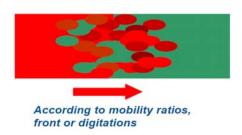
▶ Thermodynamic conditions during gas oil displacement

- Immiscible
- Oil Swelling
- Partial miscible: vaporizing gas drive or condensing gas drive
- Totally miscible at first contact

Conditions : secondary or tertiary conditions

Gas injection: main mechanisms





the reservoir fluid

- According to gas and oil composition and also reservoir pressure and temperature:
 - Miscibility is possible: single fluid obtained
 - No miscibility two phases: sweeping 2. efficiency depends on mobility ratio.



Miscibility and Miscible Floods

For a given temperature, miscibility depends on the fluid composition and the pressure:

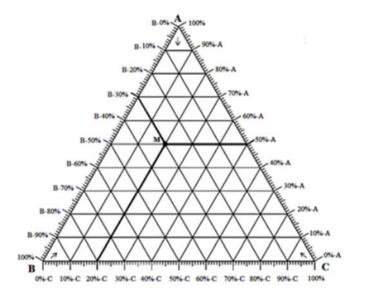
- one hydrocarbon phase → the injection process is miscible
- two separate oil and gas phases → the injection process is immiscible

In a reservoir with a GOC, there is an interface between the gas phase and the oil phase. This interface is associated to interfacial tension (IFT). As the IFT reduces to zero, the interface disappears and the two fluids become one: we have miscibility.

The main advantage of miscibility is that there is no residual oil to gas displacement. Achieving miscibility means increasing recovery.

Some definitions: the ternary diagram

Ternary or triangular phase diagrams can be used to plot the phase behavior of systems consisting of three components by outlining the composition regions on the plot where different phases exist.



Component	Weight% in mixtur (M)
A	50
В	30
C	20

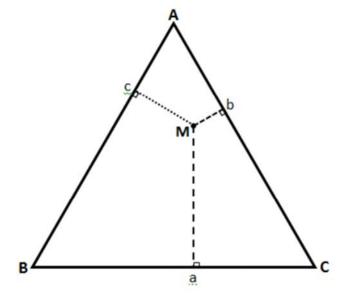


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Miscible gas injection

Some definitions: the ternary diagram

The advantage of using a ternary plot to depict compositions is that three variables can be conveniently plotted in a two-dimensional graph, and the mixture of different components can be easily represented. A ternary diagram for the hypothetical components A, B and C is:



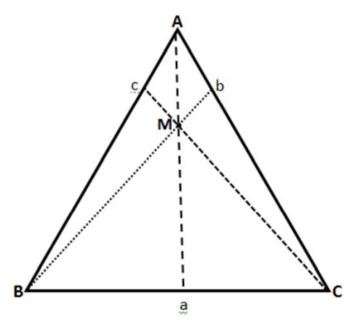
Component Weight% in mixture (M)

A
$$\frac{\overline{Ma}}{\overline{Ma} + \overline{Mb} + \overline{Mc}}$$

B $\frac{Mb}{\overline{Ma} + \overline{Mb} + \overline{Mc}}$

C $\frac{\overline{Mc}}{\overline{Ma} + \overline{Mb} + \overline{Mc}}$

Some definitions: the ternary diagram



Component	Weight% in mixture(M)
Α	$\frac{\overline{Ma}}{\overline{Aa}}$
В	$\frac{\overline{Mb}}{\overline{Bb}}$
С	$\frac{\overline{Mc}}{\overline{Cc}}$

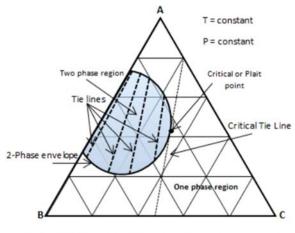


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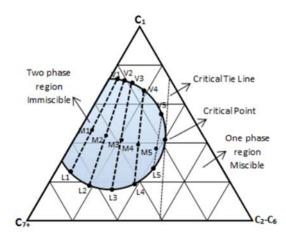
Miscible gas injection

Some definitions: the ternary diagram

Within the two-phase region there are tie lines whose ends represent the composition of equilibrium phases. The length of the tie lines shrinks toward the critical (plait) point where the properties of two phases are indistinguishable. The position of the plait point changes with temperature at a fixed pressure. Any composition represented by points (M1-M5) inside the two-phase envelop would separate into two phases (V1-V5 as vapor and L1-L5 as liquid), the relative amount of two phases can be calculated by using inverse-lever-arm rule. The points outside the two-phase envelop are representative of a single phase composition. The critical tie line is the fictitious tie line tangent to the bimodal curve at the critical point. The critical tie line is the limiting case of the actual tie line as the plait point is approached.



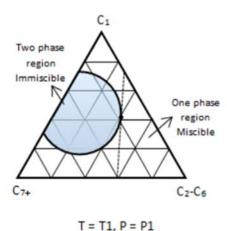
(a). ternary phase diagram for a system of components A, B, C with limited miscibility

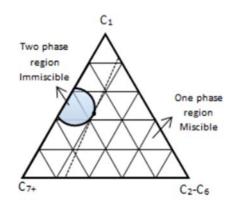


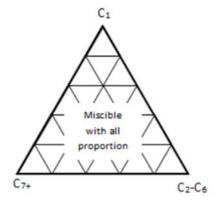
(b). Pseudoternary diagram

Some definitions: the ternary diagram

As the pressure increases, the two-phase region shrinks, or in other words light-heavy miscibility increases. No general statement is possible about the effect of temperature though the two-phase region generally grows with the increasing temperature.







$$T = T1, P = P2 > P1$$

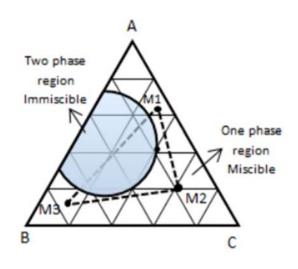
T = T1, P = P3 > P2



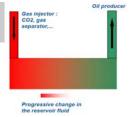
Miscible gas injection

Some definitions: the ternary diagram

In ternary diagram the mixture results of any combination of two components will lie on a straight line connecting two components to each other. According to this, any combination of components A and C, and any combination of B and C, form a single phase, in other words in this specific pressure and temperature, A and C, B and C are miscible. A and B are not miscible because the straight line between them pass through the two-phase region, so mixing them with special compositions will end in a two phase mixture. And with the same manner M1 and M2 are miscible the same as M2 and M3. But M3 and M1 are not miscible, because the straight line between them passes through the two-phase region



Compositional effects (miscibility)



Fluid characteristics

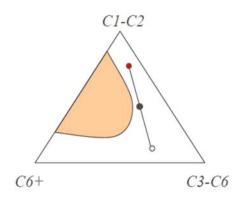
- Fluid in place: volatile oil (rich in C3-C6)
- Injected fluid: rich gas (rich in C3-C6)

Compositional effect

- Bilateral exchange of components
- No more interface between gas and oil; no interfacial tension

Consequences

- The mixture gives a single fluid
- No more Kr and Pc
- Improvement of the microscopic recovery.





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Miscible gas injection

Compositional effects (oil stripping)

Fluid characteristics

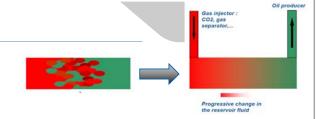
- Fluid in place: volatile oil (rich in C3-C6)
- Injected fluid: dry gas (poor in C3-C6)

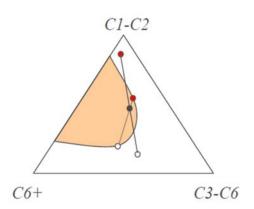
Compositional effect

• Intermediate components (C3-C6) pass from the oil into the gas

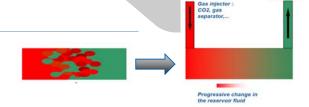
Consequences

 Possibility to recover intermediate components with gas cycling.





Compositional effects (oil swelling)



Fluid Characteristics

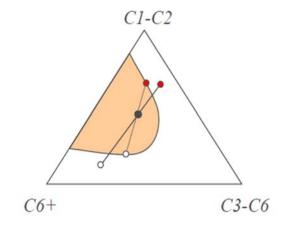
- Fluid in place: medium/heavy oil (poor in C3-C6)
- Injected fluid: rich gas (rich in C3-C6)

Compositional effect

 Intermediate components (C3-C6) pass from the gas into the oil

Consequences

- Decrease oil viscosity and density
- Enhancement of W/O mobility ratio





Progressive change in the reservoir fluid

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Miscible gas injection

Relative permeabilities

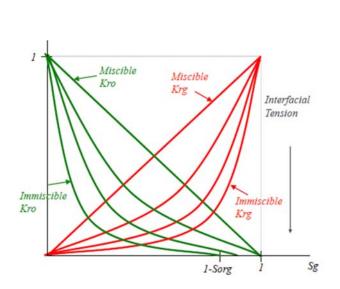
Characteristics

- Bilateral exchange of components
- No more interface between gas and oil
- No interfacial tension (IFT = 0)

Consequences

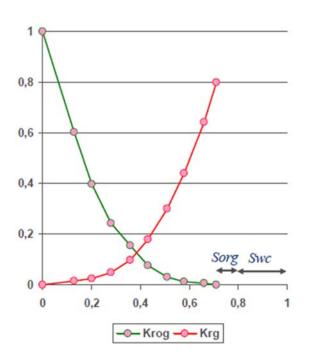
- The mixture gives a single fluid
- No more relative permeability and capillary pressure in the G/O system

Improvement of the microscopic recovery

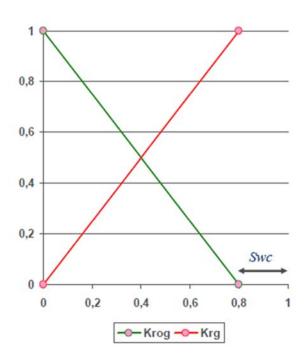


Relative permeabilities





G/O Rel. Perm (Miscible flow)



Sorg = 0 and high oil & gas mobility when flow is miscible



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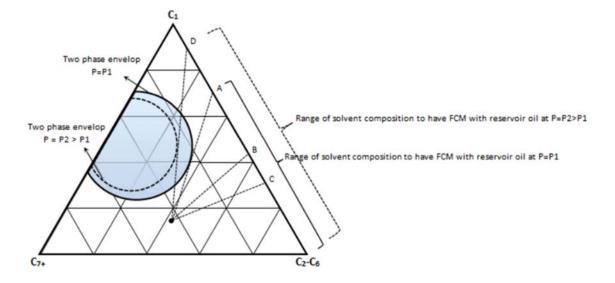
How do we get miscibility?

There is a minimum pressure required to achieve miscibility, and this pressure depends on the process. There are two main miscible processes:

- ► First-contact miscibility
- ► Multi-contact miscibility
 - Condensing drive
 - Vaporizing drive

First-contact miscibility

- ▶ In theory, first-contact miscibility can be achieved with most gases, but it crucially depends on the pressure being high enough.
- ► The first contact miscibility pressure at which any mixture of the original reservoir oil and injection gas is single phase.
- ▶ Pressures are generally too low for first-contact miscibility.

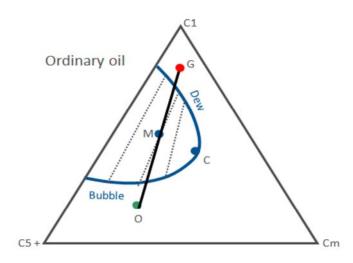


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First-contact miscibility

Composition effect



Light oil

Bubble

Cr

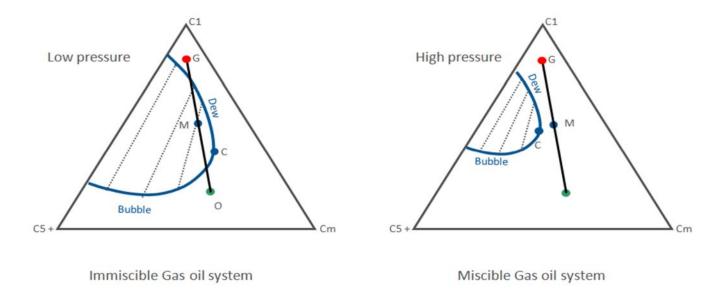
Immiscible Gas oil system

Miscible Gas oil system

Light oil is favorable to miscible gas injection

First-contact miscibility

Pressure effect

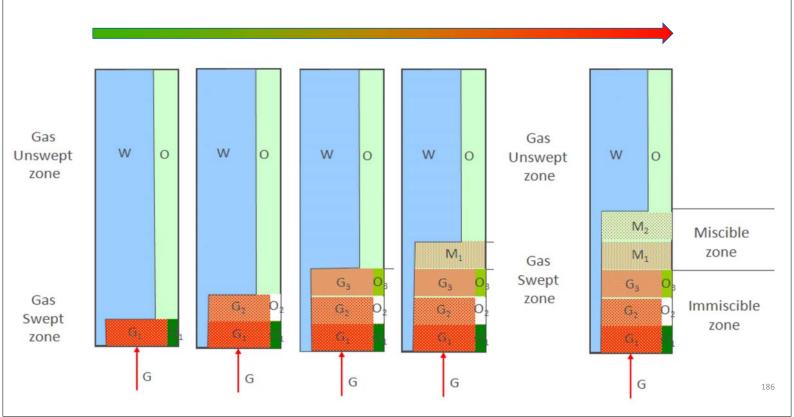


High pressure is favorable to miscible gas injection

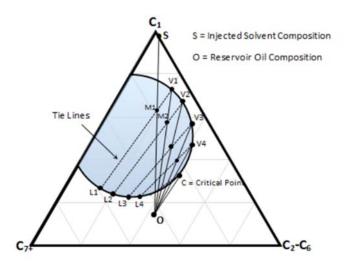


Multi-contact miscibility

For reservoirs with an initial pressure below the FCMP, we can look for an alternative process that can also give us miscibility at a Multi-contact miscibility pressure.



The injected gas 'S' after contacting the oil 'O' forms a mixture 'M1' that is split into two equilibrated phases of liquid L1 and gas V1, determined by the equilibrium tie line. It should be mentioned that the gas phase, V1, is the original solvent gas, S, after it has been enriched with some intermediate and heavy fractions from the oil phase. The gas V1 will have much higher mobility than L1 and moves forward and makes further contact with fresh oil to form mixture M2. The mixture M2 splits into gas V2 and liquid L2. The gas V2 is richer particularly in the intermediates. For the next time V2 passes L2 because of higher mobility and contacts to the fresh oil to form mixture M3 that is split into L3 and V3, and so far.



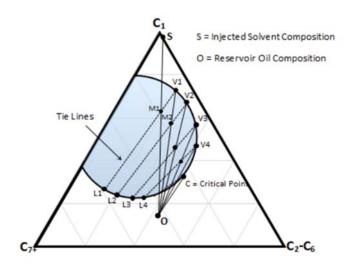


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Vaporizing gas drive miscibility

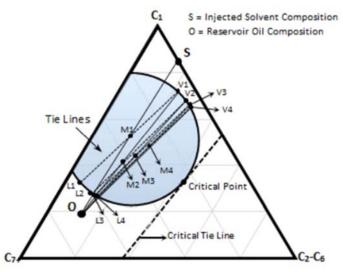
After some steps the gas phase will no longer form two phases when in contact with fresh oil. In other words the dilution straight line between 'O' and the gas phase does not pass through the two-phase region and the gas becomes miscible with oil at point 'C', that is, where the tangent line at the critical point, which is the critical tie line with zero length, goes through the oil composition 'O'.

In the vaporizing gas drive there is a transition zone; the miscibility is achieved at the front of the advancing gas; the gas composition varies gradually from that of the injected gas till reaching the 'critical point composition'. Then its miscibility displaces the original reservoir oil in a piston-type manner. No phase boundary exists within the transition zone.



Immiscible displacement

The injection gas 'S' does not achieve multiple contact miscibility with oil 'O'. The initial mixture 'M1' is the first mixture after the contact of the gas 'S' and oil 'O'. The mixture is split into the gas V1 and the liquid L1. The gas phase will flow forward to form the mixture M2, and so forth. This gas is being enriched in intermediate components at the leading edge of the solvent-oil mixing zone as discussed before. But enrichment cannot proceed beyond the gas-phase composition given by the tie line whose extension passes through the oil 'O' which is called Limiting tie line. In other words, enrichment of the advancing gas is limited by the tie line (V4-L4 here) which, if extended, goes through oil 'O'.

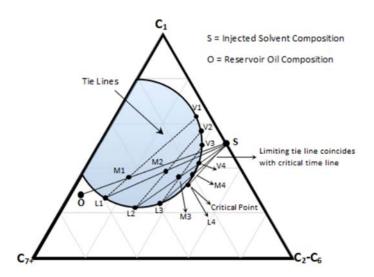




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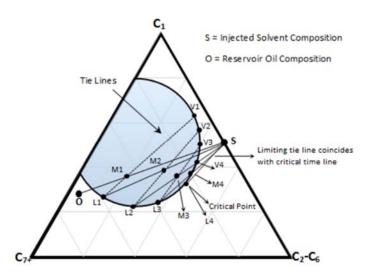
Condensing gas drive miscibility

Oil 'O' and injected fluid 'S' are not miscible initially as the dilution straight line between them passes through the two-phase region. M1 is the first mixture resulting after first contact of 'S' and 'O'. M1 will split into the liquid L1 and gas V1 that are in equilibrium at this point in the reservoir. The liquid phase L1 is richer in intermediate components than the original oil 'O'. The gas phase V1 moves faster because of its higher mobility and leaves the oil phase L1 to mix with the fresh fluid injected 'S' to form the mixture M2. The new mixture will split into liquid L2 and gas V2. The liquid L2 lies closer to the critical (plait) point than L1 and it is richer in intermediate components. The gas passes the liquid phase and L2 contact with the fresh solvent to form M3 and so forth.



By continuing the injection of the solvent 'S' the composition of the liquid phase is altered progressively in a similar manner along the bubble point curve until it reached the critical point.

The plait point fluid is directly miscible with the injection fluid 'S'. The limiting tie line in this process passes through the solvent composition 'S', so the MMP in this process is defined as the pressure at which the critical tie line coincides with the limiting tie line and its extension passes through the solvent composition



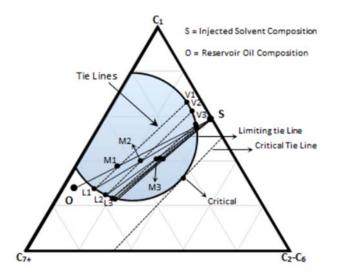


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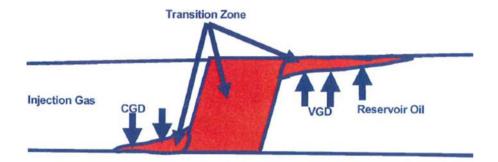
Condensing gas drive

Immiscible displacement

To achieve dynamic miscibility with the condensing-gas drive method with an oil whose composition lies to the left of the critical tie line, the enriched gas composition must lie to the right of the critical tie line. If a gas injected contains less intermediate hydrocarbon so that both oil and solvent compositions located on the two-phase side of the critical tie line the oil cannot be enriched to the point of miscibility. The enrichment of the liquid phases (L1, L2,...) continues till a point that the resulting mixture lies on the tie line that passes through the injected solvent composition point 'S'. The enrichment will stop at this point. For this system, miscibility can be achieved by increasing pressure to shrink the phase envelop, so that the limiting tie line coincides with the critical tie line



- ▶ The process that occurs in a reservoir cannot be represented as either a vaporizing or a condensing process only. Both probably take place at the same time:
 - Injection gas enriches the oil in the light intermediate range
 - Also, it strips the heavier fractions
 - Thus, the reservoir oil in contact with fresh gas initially becomes lighter, but as it contacts more gas and looses the middle intermediates and lighter heaviers, it tends to get heavier.
 - This heavier oil becomes less miscible with the injection gas.

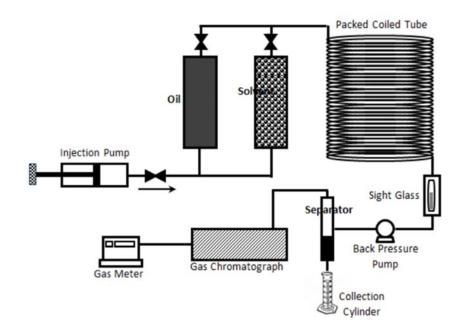




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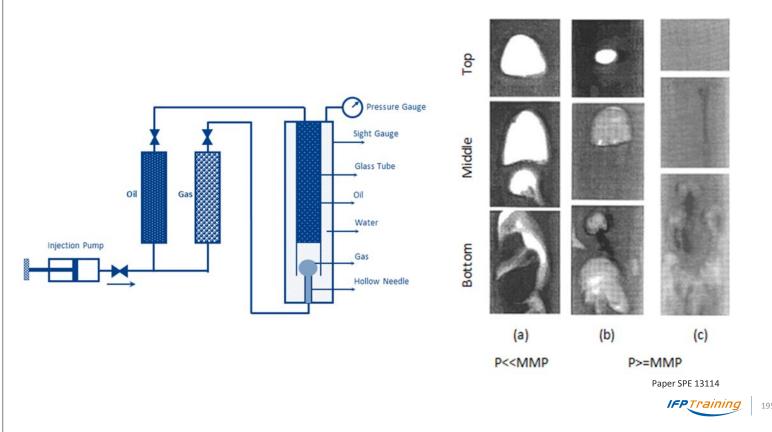
Minimum miscibility pressure (MMP): Slim tube test

Laboratory experiments are carried out to estimate the MMP. A slim tube experiment is conducted by injecting gas of a fixed composition into oil, at a number of different pressures. The oil recovery is plotted as a function of pressure. At some pressure the injected gas is to the right of the limiting tie line and MCM develops.



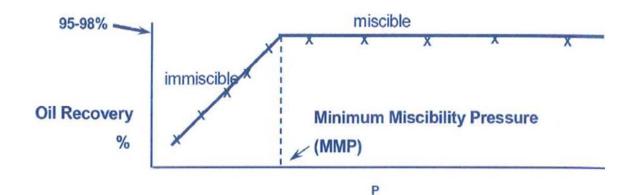
Minimum miscibility pressure (MMP): RBA

► Rising bubble apparatus (RBA)



Minimum miscibility pressure (MMP): Slim tube test

As the pressure is increased, the two-phase region becomes smaller until we reach the miscibility pressure. At that pressure the oil recovery will be maximum as the oil and gas will form a single phase throughout the slim tube, and residual oil will drop close to zero. The pressure at which the maximum recovery is first reached is the MMP.

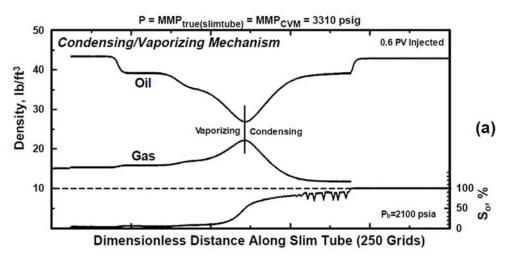


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at MMP for condensing vaporizing

A simple way to establish if a miscible slim tube simulation encounters a vaporizing or a mixed condensing/vaporizing mechanism can be seen from a plot of gas/oil densities and oil saturation versus distance along the slim tube (prior to breakthrough, e.g. at 0.6 PV injected).

An "hour-glass" shape on the density-distance plot indicates a mixed condensing/vaporizing mechanism, with the miscible front being located at the minimum in density difference. Furthermore, two phases are found on both sides of the front. The extent of the two-phase region ahead of the front may vary from very short (for a highly undersaturated system) to quite long for a slightly-undersaturated (or initially two-phase) system.



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Vaporizing gas drive miscibility

In practice...

- Multiple contact miscibility
- ▶ Lean (Separator) gas (75 to 100% C1) = continuous injection
 - 60 to 100% HCPV (10-15 years) (Prod Gas Re-injected)
- ▶ C2-C6 Transfered from oil (Light Oil ~ 40° API) to gas
- ▶ Operating pressure → 4500 psi
- Projects
 - Large scale Long period
 - Mainly secondary recovery
 - Recovery > 50% OOIP
 - Examples: Hassi Messaoud (Algeria)

Condensing gas drive miscibility

In practice...

- ► Multiple contact miscibility (C2+ transferred from Gas to Oil)
- Enriched gasi
- ▶ Slug size = 10 to 20% HCPV
- ► Operating pressure = 1,500 to 3,000 psi (for 30° API)
- Projects
 - Secondary projects mainly (Oil Gravity = 30 to 50° API)
 - Several CGD in Pinnacle Reefs (Canada)
 - Examples: Rainbow (Alberta), Intisar (Libya)
 - Estimated incremental recovery: + 15 to 25% OOIP



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Tertiary Oil Recovery by Gas Injection

- ▶ Gas injection to sweep oil zones unswept by water Water Alternate Gas
- ► Gas gravity displacement

Water alternate gas (WAG)

With miscible gas

Principles

- To optimize the microscopic recovery (gas-oil displacement)
- To optimize the volumetric recovery (water-oil displacement)

Advantages

- Management of associated gas
- Reduces gas mobility
- Sweeps zones that are not flooded by water
- Supposed to improve the microscopic efficiency

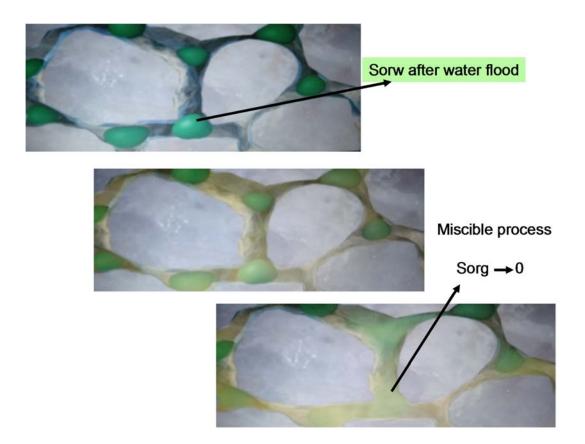
Disadvantages

- "Unnatural injection"
 - Risk of rapid fluid segregation
 - Strong sensitivity to heterogeneities
 - Risk of rapid gas breakthrough
- Decrease in water injectivity



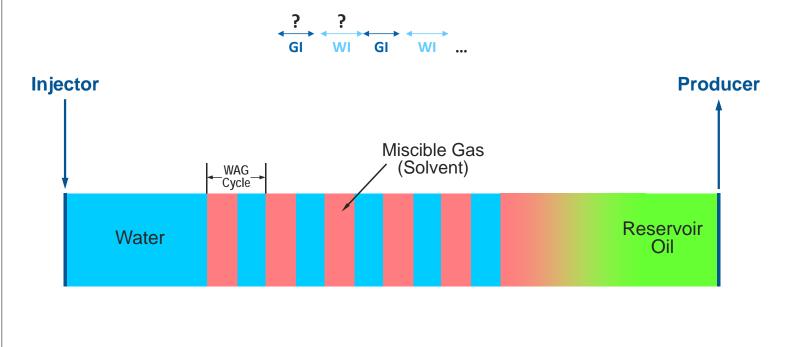
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Miscible gas injection to remove Sorw after waterflood



WAG principles

Cycles of gas and water injection alternated (injection period variable ?)

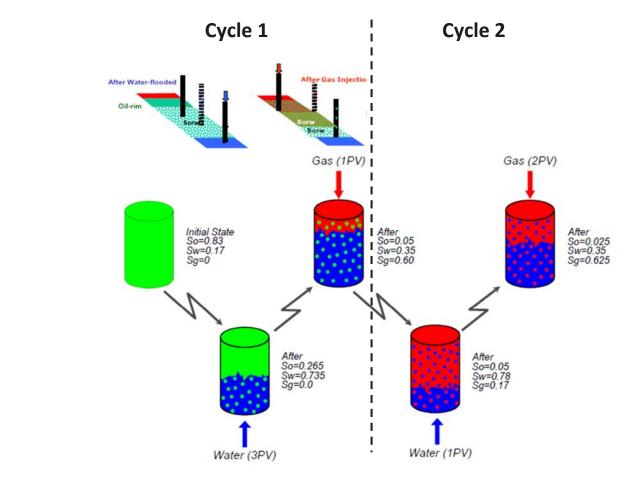




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Water Alternate Gas

2 WAG cycles (P > MMP)



Water alternate gas

Important parameters

- Reservoir thickness
- ▶ Kv, permeability barriers:
 - Works best in a stratified reservoir
- Spacing between wells
- Kh anisotropy
- ▶ WAG ratio and size of slugs



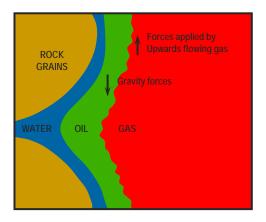
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Gas injection after waterflood

Advantages

- Residual oil recovery can be obtained
 - Either by immiscible displacement (which is beneficial if Sorg < Sorw)
 - Or through compositional effects (which can leave very low oil saturations by vaporization, or even fully sweep if strict miscibility is achieved).
- ▶ Oil sweep oil in areas not been reached by water can be obtained
 - By differences in densities
 - And possibly by new injection points.

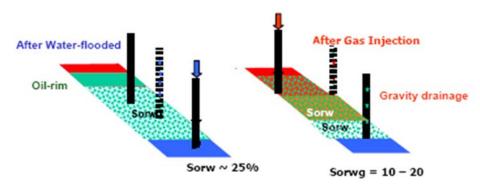
- When gas is injected at the crest of the structure, preferably in a gas cap, and if the dip angle is sufficient, the difference in gravity between gas and oil will promote segregation between the two fluids, this will allow the GOC to move downwards in gravity stable manner, despite the gas high mobility.
- ▶ Indeed, what happens is that the oil droplets will congregate in the presence of gas and will form a continuous phase that will keep on decanting slowly downwards to the producing wells.



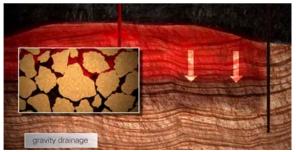


Gravity drainage effects

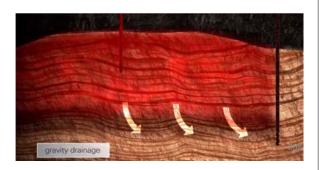
► An oil saturation of around 15-20% remains behind the gas front. This residual oil saturation will decrease with time.







▶ This is the most efficient IMMISCIBLE GAS DISPLACEMENT process.





Miscible Gas Injection

Principles

- Two fluids are miscible if they can mix in all proportions and form a single homogeneous phase
- No more interfacial tension: S_{org} tends to zero

Definitions

- Minimum miscibility pressure (MMP): the minimum miscibility pressure is the lowest pressure at which miscibility (direct or multiple contact) can be achieved, at given temperature and composition
- MMP tests: Slim tube test and Rising bubble test
- First contact miscibility: the first contact miscibility pressure at which any mixture of the original reservoir oil and injection gas is single phase.
- Multiple contact miscibility: multiple-contact miscibility: the injected gas and the insitu oil exchange components until miscibility between the two phases is reached



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Key points to keep in mind



Miscible Gas Injection

▶ Thermodynamic conditions during gas oil displacement

- Immiscible
- Oil Swelling
- Partial miscible: vaporizing gas drive or Condensing gas drive
- Totally miscible at first contact

Multiple contact miscibility mechanisms

- Condensing miscibility: intermediate components (C3-C6) pass from the gas into the oil → the mixture gives a single fluid
- Vaporizing miscibility: intermediate components (C3-C6) pass from the oil into the gas → the mixture gives a single fluid
- Condensing-vaporizing: injection gas enriches the oil in the light intermediate range and strips the heavier fractions from the reservoir oil, then gas becomes heavier and oil becomes lighter → the mixture gives a single fluid



Miscible gas injection

Advantages

- Good microscopic recovery: low residual oil saturation and interfacial tension
- Good volumetric efficiency if miscible displacement and gravity stable displacement
- Phase behavior: low viscosity, high relative permeability
- Then high injectivity
- Thermodynamic exchanges: oil swelling and miscibility

Drawbacks

- High sensitivity to gas sweep
- Unfavorable mobility ratio if unstable displacement → poor sweep efficiency
- High compression cost
- Gas availability



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Key points to keep in mind



Water alternate gas

Principles: alternated cycles of water and miscible or immiscible gas injection

- To improve miscible gas flooding stability
- To optimize the microscopic recovery (gas-oil displacement)
- To optimize the volumetric recovery (water-oil displacement)

Advantages

- Residual oil recovery can be obtained by immiscible displacement or through compositional effects (which can leave very low oil saturations by vaporization, or even fully sweep if strict miscibility is achieved). Improves the microscopic efficiency
- Oil Sweep oil in areas that have not been reached by water can be obtained by differences in densities and possibly by new injection points.
- Management of associated gas

Drawbacks

- Risk of rapid fluid segregation
- Strong sensitivity to heterogeneities
- Risk of rapid gas breakthrough
- Decrease in water injectivity



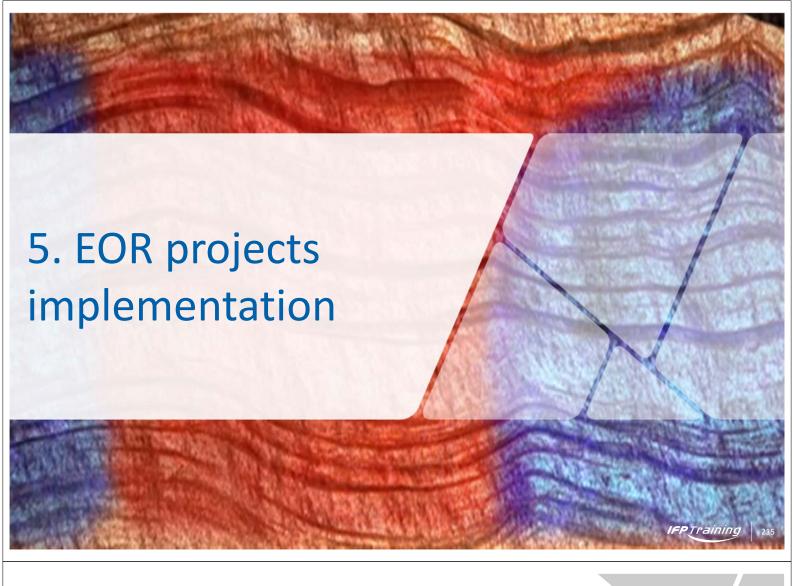
Gravity drainage

- ► Immiscible gas injection
- ▶ When gas is injected at the crest of the structure, preferably in a gas cap, and if the dip angle is sufficient, the difference in gravity between gas and oil will promote segregation between the two fluids, this will allow the GOC to move downwards in gravity stable manner, despite the gas high mobility.

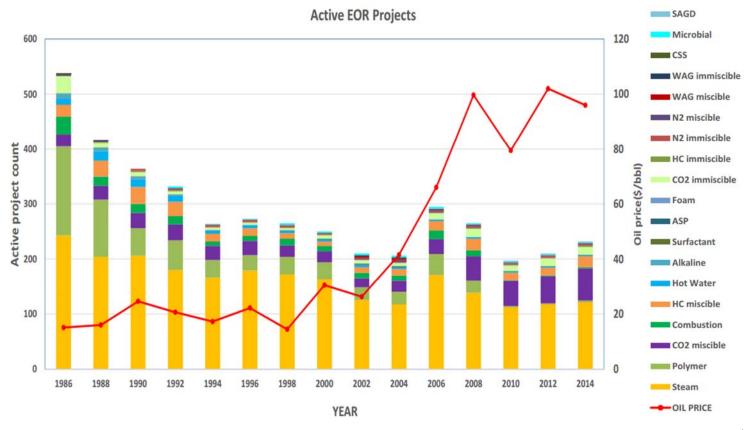


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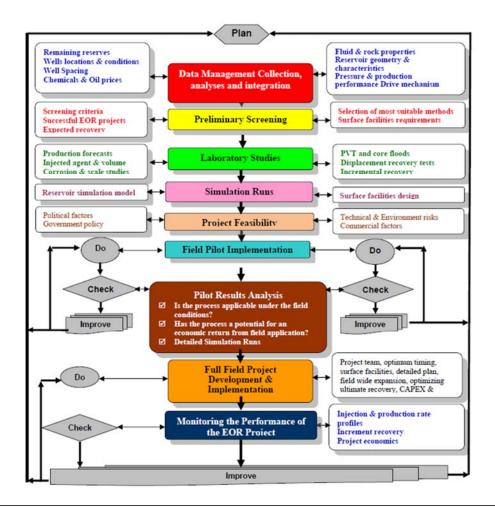
Notes



EOR project history



Design and implementation steps of an EOR program





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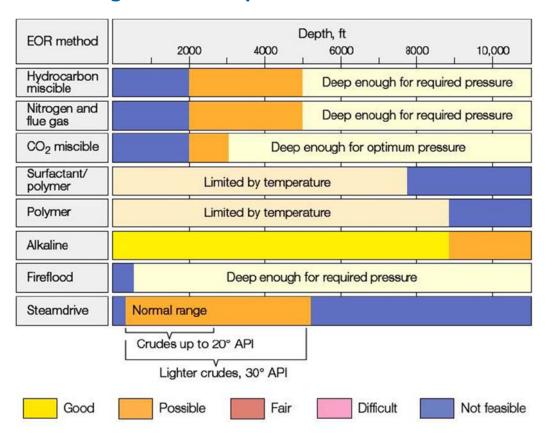
Screening

Overall screening workflow



Screening

EOR process screening criteria – depth

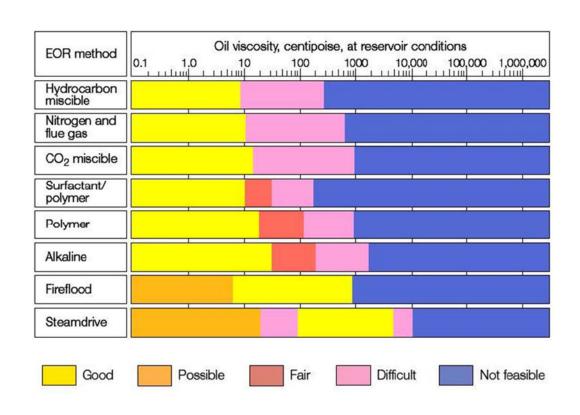




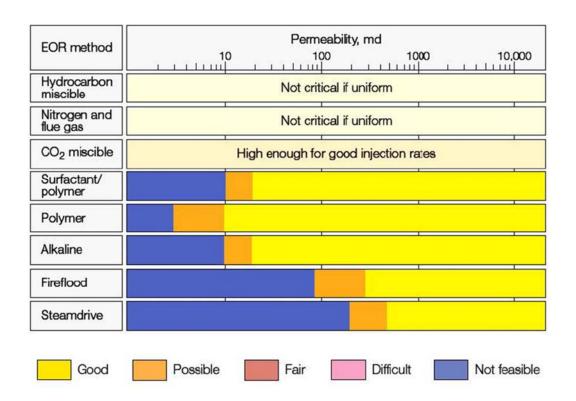
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Screening

EOR process screening criteria – viscosity



EOR process screening criteria – permeability





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Laboratory studies

- ▶ Based on the results of the preliminary screening study, several laboratory studies should be run to evaluate and assess the performance of the selected EOR method.
- ▶ The laboratory studies are undertaken to estimate incremental oil recovery as a function of the injected volume and other process variables.
- ▶ The relative permeability and displacement recovery tests on representative cores are conducted under simulated reservoir conditions.
- ► The physical properties of the system as related to injected agents are also established.
- ► The expected corrosion and scale problems to the surface piping and equipment should be carefully studied and analyzed.

The reservoir simulation model, capable of mimicking reservoir dynamics as well as the chemical/physical interaction between injection fluid and oil, is a must for planning the full pilot and field project. In this phase, all EOR project design parameters as injectors/producers pattern - type of injection - pilot plant design, etc. are highlighted. At the end of this phase, the production performance for several scenarios will guide to expect and identify the optimal hydrocarbon recovery and the required production facilities such as pipelines, processing units and storage tanks.



Project feasibility

- ▶ Selecting the best and the most feasible EOR technique to improve the recovery is carried out by comparing the net revenue of each proposed scenario.
- ▶ The net revenue for each proposed scenario depends on
 - the production performance and
 - the economic evaluation. 2.
- Potential technical risks along with the commercial factors (sources of capital; economic selection criteria; market availability; price of oil; risk tolerance); political factors (economic climatic; issues of safety/security/stability; manpower and technology availability); and government policy (long-term focus; short-term objectives; conservation; posterity concerns; and; employment focus) are also identified and studied.
- ▶ Then, the necessary recommendations for pilot field application are drawn.

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The EOR technique should be tailored to the reservoir to ensure the efficiency of the selected EOR method in the field. It is specific for a specific reservoir. Field pilot is designed and conducted as a small scale project.



Full field project development and implementation

- ► Field wide expansion follows a successful pilot project.
- ▶ This is the maturity phase of the project. In this phase, the detailed plan to maximize efficient production from the field while optimizing ultimate recovery in a practical timeframe is developed and implemented.
- ▶ The following design parameters are considered:
 - Location of the existing facilities
 - Capacity and process description of the existing plant
 - Layout of the area to define the accessibility of the future expansion of the existing facilities
 - Existing pipeline: size, length, maximum and minimum flow rate, battery limits, life time, turn down ratio of the existing pipeline
 - The existing utilities: source of the power generation and the maximum producing power
 - The available capacities of the existing storage tanks

Pilot performance monitoring and

detailed simulation studies

The surface and subsurface design parameters, along with the pilot response results are monitored, studied, analyzed and interpreted. The pilot provides much needed information for the final design and provides results that are very useful in fine – tuning of the reservoir simulation model. Then, the technical and commercial evaluation for the full field development plan is carried out. The technical evaluation will consider the following: existing wells location, fluid distribution, chemical requirements, surface facility requirements, preliminary sizing of the required equipment, associated risk, etc.



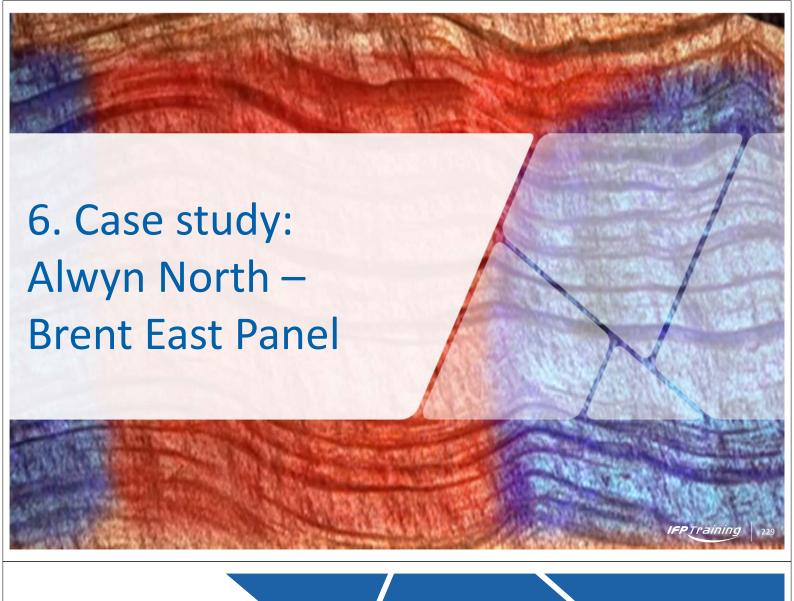
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Key points to keep in mind



Design and implementation steps of an EOR project:

- ▶ Data collection, analysis and integration
- Screening
- Laboratory studies
- ► Simulation runs (segment and full field models)
- Project feasibility
- Field pilot implementation
- Full field project development
- ▶ Monitoring the performance of EOR project



Case study outline

- ▶ Field development history
- Screening
- **▶** Slim tube simulations
- ▶ Simulation studies in segment model Miscible Gas Injection WAG
- Reservoir monitoring philosophy
- ► Early field performance of gas injection
- Reservoir management
- ► Follow up simulation studies
- Conclusions

- Sequence of sand/silts and shales deposited in deltaic and shoreface environments, $\phi \sim 18\%$ and K = 10 to 800 mD
- Offshore
- First oil: November 1987
- Water injection in early 1988
- MGI was sanctioned in 1997 and first injection was in December 1999. The MGI project was sanctioned including a waterflood phase post MGI and field depressurization, and was evaluated against a continued waterflood scenario



Screening

- Reservoir depth ~ ,m TVDSS ~ 10,500 ft TVDSS
- Formations: Tarbert and Ness
 - Upper Tabert: massive sand with good connectivity and permeability
 - Lower Tarbert: poorer quality and more channel like, locally can be good channel sands
 - Ness: channel system with poor vertical connectivity between layers
- ► Light oil: ~ 39 °API
- Reservoir temperature: 110 °C
- Initial reservoir pressure: 446 bar
- Saturation pressure at 110 °C: 257 bar → undersaturated oil
- ▶ PVT properties @ initial reservoir conditions:
 - Bo = 1.64
 - Rs = 196
 - $\mu = 0.42 \text{ cP}$

Screening

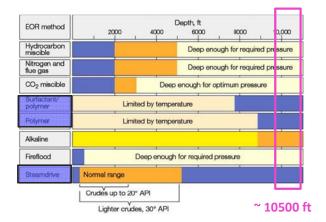
Good

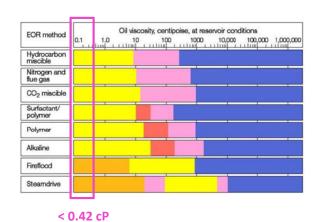
Possible

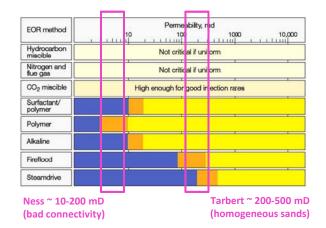
Fair

Difficult

Not feasible







Mobility:

 $\lambda o = kro/\mu o = 1.9$ $\lambda w = krw/\mu w = 1.1$

Mobility ratio:

 $M = \lambda w / \lambda o = 0.58$



, | 2

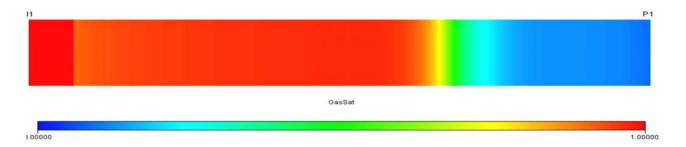
Screening

- ▶ Miscible gas injection recommended
- ▶ Separator gas and Statfjord gas available



Studies on MMP

- ► The purpose of slim tube simulation is to investigate the miscibility pressure (MMP) between the reservoir oil and the injection gases. By knowing the MMP, we can ensure that pressure will be maintained above MMP for gas injection process.
- ▶ Slim tube (1D model) used in this case is composed of 500 grid cells with 1m of length in X direction and 10m in Y and Z direction. Injector is located in the first cell (1, 1, 1) and producer is in the last cell (500,1,1).





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MMP determination

- ▶ EOS was generated using the appropriate laboratory experiments.
- ▶ The fluid in place initially is only oil phase, no water present in slim tube.
- ► MMP between oil-separator gas and oil-lean gas has already determined. The simulation will only investigate MMP of oil-rich gas and oil-CO₂.

▶ A very high permeability of the slim tube is used in order to enhance mixing of the gas and the oil and also to reduce the effect of the viscosity gradient.

Parameter			Comment
Length of Tube	500	m	500 cells of 1 m length each
Porosity	0,3	Fraction	
Pore Volume	15000	RM3	Total Pore volume from .PRT file
Absolute Permeability	50000	mD	
Control Mode		ВНР	



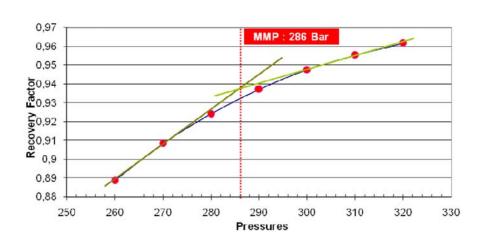
MMP determination

CO2 injection

The simulation of CO_2 injection was ran for different pressures ranging from 260 to 320 bar. The table below shows the oil in place (FOIP), the accumulative production (FOPT) and the recovery factor (RF) as a result of injecting 1.2 pore volume (PV) of CO_2 at pressures indicated

Total Pore Volume	15000	RM3
1,2 PV	18000	RM3
Injection Rate	10	RM3/d
Length of injection	1800	Days

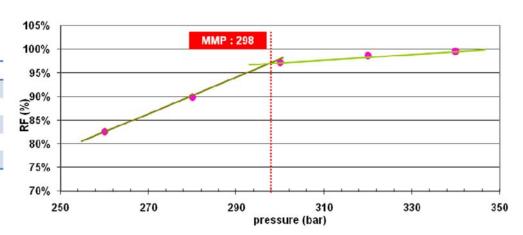
Pressure	FOIP	FOPT	RF	
260	8594.55	7638.07	89%	
270	8625.32	7834.70	91%	
280	8655.27	7998.25	92%	
290	8684.43	8138.76	94%	
300	8712.85	8256.22	95%	
310	8740.56	8349.79	96%	
320	8767.61	8431.46	96%	



Rich gas injection

The simulation of rich gas injection was ran for different pressures ranging from 260 to 340 bar. The table below shows the initial oil in place (FOIP), the accumulative production (FOPT) and the recovery factor (RF) as a result of injecting 1.2 pore volume (PV) of rich gas at pressures indicated,

Pressure	OOIP	FOPT	RF
260	8594.55	7089.23	82%
280	8655.27	7778.7	90%
300	8712.85	8464.73	97%
320	8767.61	8647.67	99%
340	8819.82	8778.08	100%





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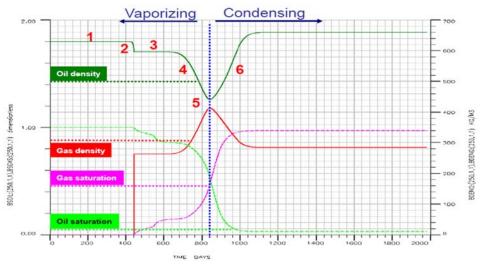
MMP comparison

- ► CO₂ has the lowest MMP compared to rich and lean gas.
- ▶ Separator gas that more or less has the same composition as lean gas has the highest MMP.
- ▶ The initial reservoir pressure is 446 bar, which is higher than the MMP of all the gases injected, thus all the gases can be used for miscible injection.
- ► To inject lean gas and separator gas, higher pressure maintenance in reservoir is required.

Rich gas injection: displacement process

Density and saturation plot of rich gas injection process at 298 bar (MMP) on grid number 250

- 1. Oil saturation (So) still at its initial because injected gas has not arrived,
- 2. After the injected gas reach grid 250, gas is dissolved in the oil and it lightens the oil (swelling) hence it reduces oil density, then
- 3. Gas will act as a free gas,
- 4. Vaporizing mechanism happens when intermediates of the oil vaporize into the gas thus density-saturation of oil decrease and density-saturation of gas increase,
- 5. Until it reaches miscibility/near miscibility. In the miscibility region, density difference between oil and gas will be very small and composition of two fluids will be more or less the same.
- 6. After miscibility occurs, condensing mechanism happens. In this process, intermediate in gas will move into oil because oil density increases and gas density decreases, but gas saturation keeps increasing because more gas injection arrives in the grid 250 and more oil is displaced by gas.

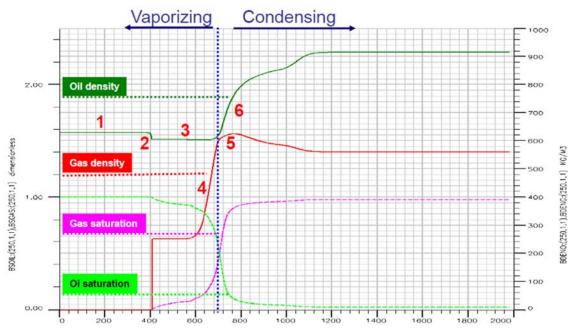


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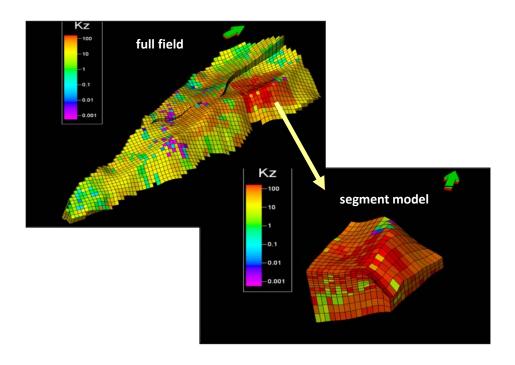
CO₂ injection: displacement process

In the density and saturation plot of CO_2 injection process at 286 bar (MMP) on grid number 250, miscibility on CO_2 occurs earlier compared with rich gas injection. The process (steps 1 to 6) is more or less the same as in rich gas injection.



TIME DAYS

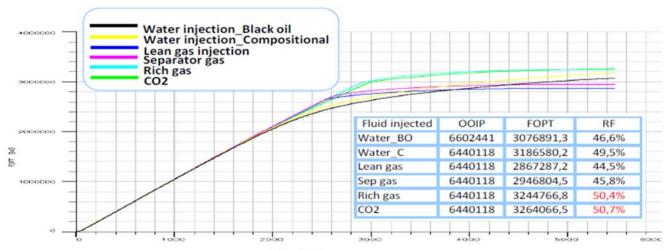
The purpose of segmenting a full field model is to evaluate a very sensitive case in a high resolution model. With a segment model, the CPU time can be reduced and a sensitivity study, which requires lots of runs can be made in a high resolution model.





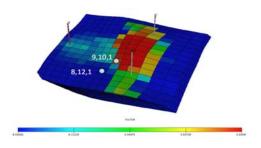
Comparison

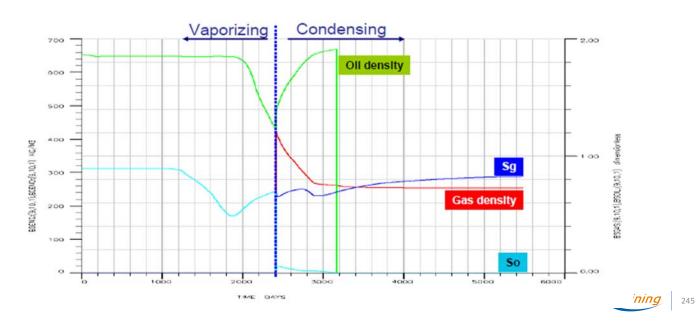
- ► CO₂ and rich gas injection have a good performance, these two gases have later gas breakthrough.
- ▶ The recovery factor (RF) for CO₂ and a rich gas are also higher.
- ► Lean gas and separator gas act as a immiscible injection, field pressure during gas injection is below their MMP.



Sweep efficiency: rich gas injection

Evidence of miscibility (9,10,1):





Water alternating gas injection

▶ The objective is to find an optimized cycle for WAG. As a basis of comparing recoveries we use additional recoveries when WAG is performed (after 4 years of water injection, same start time). In this case the WAG ratio is almost the same (1:1).





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Reservoir monitoring philosophy (1/2)

An intensive reservoir monitoring is carried out to ensure that MMP is maintained and to identify performance changes

- ▶ Downhole pressure data and surveys: permanent gauges in four key wells. As one well (N10) has been shut for over a year, constant static data have been acquired
- ► Wellhead monitoring: WHFP, WHFT & BSW are daily monitored. WHFP is a key sign of gas breakthrough
- ▶ Well tests: undertaken on a two-month basis. The trend of oil potential, GOR and WCT evolution are closely monitored.
- ► Flow profiles PLT data: PLT were run in all the producers before gas injection to provide a baseline and they are run at any change in the behavior of a well (gas breakthrough, thickness of the gas front for sweep efficiency, injectivity indicators, etc...)

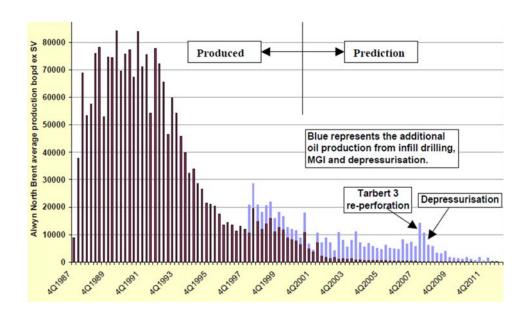
- ► Fluid saturations RST data: RST were run in 5 producers before gas injection to provide a baseline; the tests indicate the degree of gas banking (no conclusive due to scale and invasion of workover fluids).
- ▶ Tracers: chemical tracers were injected into each injector at the start of gas injection. Periodic samples have been taken from all the producers for analysis to detect the tracer. These results can be used to help determine the sweep efficiency of the injected gas by means of time of flight versus distance between injector/producers.
- ▶ Fluid composition: through fluid sampling. Lighter produced fluid has been observed due to the exchange of components between the reservoir oil and the injected gas.
- ► Injectivity tests: injectivity tests were performed soon after the first gas injection and in all the cases injectivity was not a problem.



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Early field performance of gas injection

▶ Incremental oil (production and prediction):



Good reservoir management is essential to a project of this nature where voidage maintenance is critical. Voidage can be severely compromised if gas cycling occurs on a large scale.

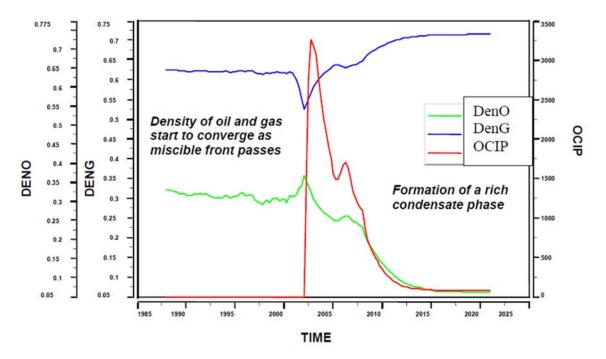
It has been estimated that by mid 2002 around 5 to 9% of the injected gas has been back produced.

This shows no major problem of gas cycling and indicates that there must be a reasonable sweep efficiency in the reservoir.



Follow up simulation studies

- Simulation modeling is used for both reservoir management near term and for long term performance predictions.
- Using the model as a tool requires the model to be regularly updated.
- Detailed simulation studies were carried out:
 - Zonal gas injection: injecting in the lower zones for a period of time, then switching to the upper zones,
 - Conversion of selected producers to injectors to help with the voidage constraints,
 - Water alternate gas (WAG),
 - Vertical effect of gravity on the macroscopic mechanisms,
 - Lateral effects such as gas conduit in the channel sands.



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Conclusions

Production strategy:

- Natural depletion
- Waterflooding for 12 years in total voidage replacement (1988 1999)
- Miscible gas injection for 5 years partially in WAG process (1999 2005)
- Blowdown production with no more pressure maintenance

Comprehensive reservoir simulation studies are key to near term reservoir management as well as long term prediction